# Relativistic all-order calculations of $\mathbf{T h}, \mathbf{T h}^{+}$, and $\mathbf{T h}^{\mathbf{2 +}}$ atomic properties 

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#### Abstract

Excitation energies, term designations, and $g$ factors of $\mathrm{Th}, \mathrm{Th}^{+}$, and $\mathrm{Th}^{2+}$ are determined using a relativistic hybrid configuration-interaction (CI) plus all-order approach that combines configuration-interaction and linearized coupled-cluster methods. The results are compared with other theoretical and experimental values where available. We find some vanishing $g$ factors, similar to those known in lanthanide spectra. Reduced matrix elements, oscillator strengths, transition rates, and lifetimes are determined for $\mathrm{Th}^{2+}$. To estimate the uncertainties of our results, we compare our values with the available experimental lifetimes for higher $5 f 7 p^{3} G_{4}, 7 s 7 p^{3} P_{0}$, $7 s 7 p^{3} P_{1}$, and $6 d 7 p^{3} F_{4}$ levels of $\mathrm{Th}^{2+}$. These calculations provide a benchmark test of the CI plus all-order method for heavy systems with several valence electrons and yield recommended values for transition rates and lifetimes of $\mathrm{Th}^{2+}$.


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## I. INTRODUCTION

The ${ }^{229} \mathrm{Th}$ nucleus provides a unique opportunity for the development of a nuclear clock [1] due to an unusually low first excitation energy of only several $\mathrm{eV}[2,3]$, making the corresponding nuclear transition accessible with laser excitation [4]. This very narrow, $6 h$ lifetime [4], nuclear transition is expected to be well isolated from effects of external fields leading to potentially very small ultimate uncertainty in the corresponding frequency standard.

The transition frequency is expected to be very sensitive to temporal variation of the fine-structure constant and the dimensionless strong interaction parameter $m_{q} / \Lambda_{\mathrm{QCD}}$ as compared to atomic transitions, making ${ }^{229} \mathrm{Th}$ one of the most attractive candidates for such studies [5]. The physical implementation of the frequency standard may employ, for example, the closed electronic shell of $\mathrm{Th}^{4+}$ in a UV-transparent crystal doped with a macroscopic number of ${ }^{229} \mathrm{Th}$ nuclei [6] or the stretched states within the $5 f_{5 / 2}$ electronic ground level of both nuclear ground and isomer manifolds of a single trapped ion [7]. Laser-cooled Wigner crystals of ${ }^{229} \mathrm{Th}^{3+}$ allow for high-precision spectroscopy $[8,9]$. Singly and doubly charged ${ }^{232} \mathrm{Th}$ and ${ }^{229} \mathrm{Th}$ ions have been produced by laser ablation of solid-state thorium compounds and by inductively coupled plasma techniques with mass-spectrometry analysis from liquid solutions of thorium [10]. The latter method was found to be more applicable for producing ions of radioactive ${ }^{229} \mathrm{Th}$ for laser experiments when searching for the energy value of the isomeric nuclear transition [10].

We note that a frequency standard based on the nuclear transition in ${ }^{229} \mathrm{Th}$ can be implemented with either neutral or ionized Th . The nuclear transition might be accessed using an electronic bridge process [11], which involves matching combined electronic and nuclear energy levels to drive the nuclear transition [11]. Implementing the electronic bridge will require knowledge of the complex electronic configurations of neutral or ionized Th.

Much of the recent work on the spectrum of thorium is motivated by its use as source of wavelength standards for high-resolution spectrographs [12]. The high density of Th I spectral lines-approximately 20000 in the wavelength interval between 250 and 5500 nm -has made the thoriumargon hollow-cathode lamp a convenient tool for accurate wavelength calibration. In particular, such lamps are installed on many high-precision astronomical spectrographs. These include the High Accuracy Radial velocity Planet Searcher instrument, now the world's most precise astronomical spectrograph, which has a relative precision of 3 parts in $10^{9}$ and can measure stellar velocities to within $\sim 1 \mathrm{~m} / \mathrm{s}$ [13].

The best characterized Th ion today is $\mathrm{Th}^{3+}$, which has a monovalent Fr-like electronic structure and has been studied with the high-precision all-order method [14]. Recommended values for electric dipole matrix elements, oscillator strengths, transition rates, lifetimes, scalar and tensor polarizabilities, and hyperfine constants critically evaluated for their accuracy have been published for a large number states in $\mathrm{Th}^{3+}$ [14]. The combination of the experimental measurements of hyperfine constants of $\mathrm{Th}^{3+}$ with theoretical calculations has enabled accurate determination of magnetic dipole and electric quadrupole moments [15]. Analysis of resonant excitation Stark ionization spectroscopy spectra has led to the determination of the $\mathrm{Th}^{3+}$ ground-state electric quadrupole moment, adiabatic scalar and tensor dipole polarizabilities, and the dipole matrix elements connecting the ground level to low-lying excited levels of $\mathrm{Th}^{3+}$ [16].

Due to the more complicated atomic structure of $\mathrm{Th}^{2+}$, $\mathrm{Th}^{+}$, and Th , which have, respectively, two, three, and four valence electrons, their atomic properties are less precisely known than that of $\mathrm{Th}^{3+}$. The work presented in this paper demonstrates accurate calculation of the energies of these systems and provides transition matrix elements, oscillator strengths, and lifetimes for a large number of $\mathrm{Th}^{2+}$ states. This work also serves as a benchmark test of the accuracy of the CI plus all-order method for systems with multiple valence electrons, including tetravalent neutral Th.

## II. REVIEW OF CURRENT KNOWLEDGE OF THE STRUCTURE OF THORIUM AND ITS IONS

Properties of thorium and its ions have received more extensive experimental investigation than theoretical, due to significant difficulties in accurate first-principles calculations of such heavy many-electron systems. We begin with a summary of the experimental work and conclude with a survey of theoretical approaches.

A list of about 9500 spectral lines of Th in the range 234.52966.2 nm was obtained and characterized by Zalubas [17]. This resulted in determination of 254 even-parity and 322 odd-parity levels. Their $g$ factors were obtained by Zeeman spectroscopy and used as an aid in spectral classification, i.e., the assignment of angular momenta, parity, and electronic configurations.

Energy levels and classified lines in the second spectrum of thorium (Th II) were described by Zalubas and Corliss [18]. About 6500 lines were classified as transitions between 199 odd levels and 271 even levels; 188 levels result from the odd $5 f 6 d^{2}+5 f 7 s^{2}$ and $6 d^{2} 7 p+7 s^{2} 7 p$ configurations [18]. The 235 levels of the even $5 f^{2} 7 s+5 f 7 s 7 p+5 f 6 d 7 p+5 f^{2} 6 d$ configurations were also determined [18]. Resonantly enhanced three-photon ionization of $\mathrm{Th}^{+}$was used to determine that its ionization potential is between 11.9 and 12.3 eV [19].

The Th III spectrum was observed in the region 10003000 nm and ten lines were identified [20]. The ground level of $\mathrm{Th}^{2+}$ was determined to be $5 f 6 d^{3} H_{4}$ [20]. The first excited level $6 d^{2}{ }^{3} F_{2}$ was determined to be only $63.2 \mathrm{~cm}^{-1}$ above the ground level. Wyart and Kaufman [21] extended the analysis of $\mathrm{Th}^{2+}$. They classified the 92 lines above 194 nm that were previously observed [21]. The $5 f 6 d, 5 f^{2}, 5 f 7 d$, and $5 f 8 s$ configurations were completely identified [21].

Using time-resolved laser-induced fluorescence method, Biémont et al. measured the lifetimes of six levels belonging to the $5 f^{2}, 5 f 7 p, 7 s 7 p$, and $6 d 7 p$ configurations of $\mathrm{Th}^{2+}$ [22]. These transitions provide the mechanism of a cosmochronometer for estimating the age of the galaxy.

An online database of published and unpublished actinide energy levels [23] lists 693 levels of Th and 507 of $\mathrm{Th}^{+}$ with uncertainties of $0.001 \mathrm{~cm}^{-1}$. Recently, Redman et al. [12] presented results of precise observations of a thorium-argon hollow-cathode lamp emission spectrum in the region between 3500 and 1175 nm using a high-resolution Fourier transform spectrometer. Their measurements were combined with results from previously published thorium line lists [24-29] to reoptimize the energy levels of neutral, singly ionized, and doubly ionized thorium ( $\mathrm{Th}, \mathrm{Th}^{+}$, and $\mathrm{Th}^{2+}$ ). A systematic analysis of previous measurements in light of these results enabled Redman et al. [12] to identify and propose corrections for systematic errors and typographical errors and incorrect classifications in previous identifications. Redman et al. [12] present 787 levels of Th I and 516 of Th II in their tables.

Dzuba and Flambaum [30] presented analytical estimates and numerical calculations showing that the energy-level density in open-shell atoms increases exponentially with excitation energy. They used the relativistic Hartree-Fock and configuration-interaction methods to calculate the densities of states of Th and $\mathrm{Th}^{+}$. Their results were used to estimate the
effect of electrons on the nuclear clock transition discussed in the previous section [30].

Porsev and Flambaum [31] used the CI plus many-body perturbation theory (MBPT) method to study the effect of atomic electrons on the nuclear clock transition due to the electronic bridge process. They calculated energies of several high-lying even-parity states that have yet to seen by experiments. Roy et al. have performed relativistic two-component $a b$ initio calculations for $\mathrm{Th}^{+}$and $\mathrm{Th}^{2+}$ ions [32].

In our present work we evaluate atomic properties of Th , $\mathrm{Th}^{+}$, and $\mathrm{Th}^{2+}$ using the CI plus all-order approach. Excitation energies and $g$ factors are compared with experimental [23] and other theoretical results [33]. We also evaluate multipole transition rates and lifetimes of low-lying levels for $\mathrm{Th}^{2+}$.

## III. COMPUTATIONAL METHOD

Calculation of the properties of thorium and it first few ions requires an accurate all-order treatment of electron correlations. Low-order perturbation methods are ineffective in heavy systems with more than one valence electron, due to large effects of valence-valence electronic correlations. Moreover, the radonlike core of the thorium atom is sufficiently large that core-core and core-valence correlations have to be treated accurately as well. This can be accomplished within the framework of the CI plus all-order method that combines configuration-interaction and coupled-cluster approaches [3438]. The CI plus all-order method was used to evaluate properties of systems with three valence electrons in Refs. [3942]. Properties of systems with four valence electrons were calculated with the CI plus all-order method in Ref. [42] for Sn -like ions. The spectra of the superheavy elements No, Lr, and Rf with two, three, and four valence electrons was recently presented by Dzuba et al. [43].

In the CI plus all-order method, we start with a solution of the Dirac-Fock (DF) equations

$$
\begin{equation*}
H_{0} \psi_{c}=\varepsilon_{c} \psi_{c}, \tag{1}
\end{equation*}
$$

where $H_{0}$ is the relativistic DF Hamiltonian [34,44] and $\psi_{c}$ and $\varepsilon_{c}$ are single-electron wave functions and energies. The calculations are carried out in the $V^{\mathrm{N}-4}, V^{\mathrm{N}-3}$, and $V^{\mathrm{N}-2}$ potentials for $\mathrm{Th}, \mathrm{Th}^{+}$, and $\mathrm{Th}^{2+}$, respectively, where $N$ is the total number of the electrons. Therefore, the calculations are carried out with the same radonlike $\mathrm{Th}^{4+}$ frozen-core DiracFock potential of all three systems considered in this work.

The wave functions and the corresponding low-lying energy levels are determined by solving the many-electron relativistic equation for two, three, or four valence electrons [45]

$$
H_{\mathrm{eff}}\left(E_{n}\right) \Phi_{n}=E_{n} \Phi_{n}
$$

The effective Hamiltonian is defined as

$$
\begin{equation*}
H_{\mathrm{eff}}(E)=H_{\mathrm{FC}}+\Sigma(E) \tag{2}
\end{equation*}
$$

where $H_{\mathrm{FC}}$ is the Hamiltonian in the frozen-core approximation. The energy-dependent effective Hamiltonian term $\Sigma(E)=\Sigma_{1}+\Sigma_{2}$ is calculated using a modified version of the all-order linearized coupled-cluster method with single and double excitations described in [46,47]. Therefore, the effective Hamiltonian contains dominant core and core-valence correlation corrections to all orders. The valence correlations
are treated by the CI method [45]. We refer the reader to Ref. [34] for the formulas and detailed description of the CI plus all-order method.

The CI plus all-order approach is based on the BrillouinWigner variant of the many-body perturbation theory, rather than the Rayleigh-Schrödinger variant, leading to dependence of $\Sigma$ upon energy. This introduces some subtleties associated with appropriate treatment of energy denominators. This issue has been discussed in Ref. [34] and we adopt the technical procedure that was recommended there.

The configurations are strongly mixed in all three systems considered here. We present the results for the following configurations: (i) Th: $6 d^{2} 7 s^{2}, 6 d^{3} 7 s, 6 d^{4}, 6 d^{2} 7 s 7 p, 5 f 6 d^{2} 7 s$, $5 f 6 d 7 s^{2}, 6 d 7 s^{2} 7 p$; (ii) $\mathrm{Th}^{+}: 6 d 7 s^{2}, 6 d^{2} 7 s, 6 d^{3}, 5 f^{2} 7 s$, $5 f^{2} 6 d, 5 f 6 d 7 s, 5 f 7 s^{2}, 5 f 6 d^{2}$; and (iii) $\mathrm{Th}^{2+}: 5 f 6 d, 5 f 7 d$, $5 f 7 s, 5 f 8 s, 6 d 6 f, 6 d 7 p, 6 d^{2}, 5 f^{2}, 7 s^{2}, 5 f 7 p, 5 f 6 f, 6 d 7 s$. We also calculate $g$ factors and compare them with experimental values given in Ref. [23]. For a single configuration that is described by pure $L S$ coupling, the nonrelativistic $g$ factor of the many-electron state is given by the Landè formula

$$
\begin{equation*}
g_{\mathrm{nr}}=1+\frac{J(J+1)-L(L+1)+S(S+1)}{2 J(J+1)} \tag{3}
\end{equation*}
$$

where $J$ is the total angular momentum, $L$ is the total orbital angular momentum, and $S$ is the total spin angular momentum. As will be discussed in the next section, we find that the nonrelativistic $g$ factors are often useful for spectral term classification.

## IV. EXCITATION ENERGIES OF $\mathbf{T h}, \mathbf{T h}^{+}$, and $\mathbf{T h}^{\mathbf{2 +}}$

Excitation energies of the 73 lowest states of neutral Th are listed in Table I. All energies are given relative to the $6 d^{2} 7 s^{2}{ }^{3} F_{2}$ ground state. Theoretical results calculated with the CI plus all-order method are listed in columns labeled "Present." The results are compared with experimental energies and $g$ factors given in Ref. [23]. Some of the energy levels listed in [23] are only identified by the total angular momentum $J$ and not by a complete $L S J$ term designation. Such designations are always approximate and sometimes ambiguous, as in cases of strong configuration mixing.

## A. Term identification using $\boldsymbol{g}$ factors

All of the states reported in Table I have some admixture of configurations, leading to ambiguities in term identification. Thus, we list two sets of term designations in Table I: the term listed in [23] and our term identification, which is based on comparing the experimental $g$ factor with the Landè formula value given by Eq. (3) as described below. The corresponding columns are labeled "[23]" and "Present." First, we group all levels by $J$ and by parity. This leads to a relatively small number of possible terms for each level since there are only three configuration present for even levels and four configurations for odd levels for the energy range in Table I. We then identify the term appropriate for each level using the agreement of the experimental $g$ factor with the Landè formula. When several configurations have the same $J$ and parity, we also verify which configuration has the largest mixing coefficient. When no entry appears in a term column that
indicates that no term has been proposed. The $g_{\mathrm{nr}}$ calculated using Eq. (3) are given in columns " $g_{\mathrm{nr}}$." The "Present" column reports the actual $g$ factor that we calculate for the state as a whole. In general, the calculated $g$ factors are in good agreement with the Landè formula, but there are exceptions in cases of strong configuration mixing.

Two interesting entries in Table I are the levels $6 d^{3} 7 s^{5} F_{1}$ and $6 d^{2} 7 s 7 p^{5} F_{1}$, for which the column " $g_{\mathrm{nr}}$ " reports a value of zero for the theoretical $g$ factor. That factor is the Landè $g$ factor given by Eq. (3), which (but for a small correction due to the anomalous magnetic moment of the electron) describes the gyromagnetic ratio of a single-configuration quantum state with well-defined quantum numbers $L, S$, and $J$ (and orbital, spin, and total electronic angular momentum, respectively, in units of the reduced Planck constant $\hbar$ ). A vanishing (or very small) $g$ factor for a level implies that its energy is relatively insensitive to the presence of a magnetic field. Such an attribute is of interest in applications to atomic frequency standards and precision measurement, so we comment briefly on the use of Eq. (3) as a screen in the search for small $g$ factors.

From Eq. (3) we find that $g=0$ when

$$
\begin{equation*}
3 J(J+1)-L(L+1)+S(S+1)=0 \tag{4}
\end{equation*}
$$

where $L \geqslant 0$ is an integer, $J$ and $S$ are both non-negative integers or half-integers, and $|L-S| \leqslant J \leqslant L+S$, the triangularity condition, is satisfied. Solution of the constrained equation (4) is an exercise in integer programming, for which even linear examples fall in the NP-hard class of computational complexity [48]. However, since only relatively small values of $J, L$, and $S$ are relevant to the Periodic Table, the practical solutions of Eq. (4) can be found by searching tables of computed values of its left-hand side.

Most of these solutions are of the type valid for any $J$ :

$$
\begin{equation*}
L=2 J+1, \quad S=J+1 \tag{5}
\end{equation*}
$$

i.e., those with term designations ${ }^{2 J+3}(2 J+1)_{J}$ for $J=$ $0, \frac{1}{2}, 1, \ldots$ Some of these terms are listed in Tables 3 and 4 of Ref. [49].

A number of such levels are identified in the NIST Atomic Spectra Database [50], where they are usually found to have $g$ factors of 0.01 or less. Examples include Nd I, $4 f^{4} 6 s^{25} \mathrm{~F}_{1}$; Pm II, $4 f^{5} 6 s^{7} \mathrm{H}_{2}^{\circ}$; Pm II, $4 f^{5} 6 s^{5} \mathrm{~F}_{1}^{\circ}$; Tb I, $4 f^{8} 5 d 6 s^{26} \mathrm{G}_{3 / 2}$; Dy II, $4 f^{10}\left({ }^{5} \mathrm{I}\right) 5 d^{6} \mathrm{G}_{3 / 2}$; Sm I, $4 f^{6} 5 d\left({ }^{8} \mathrm{H}\right) 6 s^{7} \mathrm{H}_{2}$; and Sm I, $4 f^{6} 5 d\left({ }^{6} \mathrm{H}\right) 6 s^{7} \mathrm{H}_{2}$.

There are other solutions of Eq. (4) for values of $J, L$, and $S$ that do not satisfy Eq. (5), such as $J=3 / 2, L=11$, and $S=21 / 2$. These all appear to be associated with highly excited states involving multiple open shells.

## B. Hyperfine Landè $g$ factors

A similar method can be used to search for vanishing Landè $g$ factors $g_{F}$ for hyperfine levels. Ignoring contributions from the nuclear magneton and the electron anomaly, we have (see Ref. [51])

$$
\begin{equation*}
g_{F}=g_{J} \frac{F(F+1)+J(J+1)-I(I+1)}{2 F(F+1)} \tag{6}
\end{equation*}
$$

where $g_{J}$ is the $g$ factor associated with the electrons, $J$ the net electronic angular momentum, $F$ the total angular

TABLE I. Levels (in $\mathrm{cm}^{-1}$ ) and $g$ factors of the lowest states of thorium. Nonrelativistic values of $g$ factors $g_{\mathrm{nr}}$ are given by Eq. (3). Calculations are compared with experiment [23]. Configuration and term labels are determined as described in Sec. IV A.

| Even-parity states ( $\left.6 d^{2} 7 s^{2}+6 d^{3} 7 s\right)$ |  |  |  |  |  |  |  |  | Odd-parity states ( $\left.6 d^{2} 7 s 7 p+5 f 6 d^{2} 7 s+5 f 6 d 7 s^{2}+6 d 7 s^{2} 7 p\right)$ |  |  |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Conf. | Term |  |  | Energy |  | $g$ factor |  |  | Conf. | Term |  |  | Energy |  | $g$ factor |  |  |
|  | [23] | Present | $J$ | Present | Expt. | Expt. | Present | $g_{\mathrm{nr}}$ |  | [23] | Present | $J$ | Present | Expt. | Expt. | Present | $g_{\text {nr }}$ |
| $6 d^{2} 7 s^{2}$ | ${ }^{3} F$ | ${ }^{3} F$ | 2 | 0 | 0 | 0.735 | 0.718 | 0.667 | $5 f 6 d 7 s^{2}$ | ${ }^{3} P$ |  | 0 | 13954 | 14247 | 0.0 | 0.000 | 0.0 |
| $6 d^{2} 7 s^{2}$ |  | ${ }^{3} P$ | 2 | 3790 | 3688 | 1.255 | 1.287 | 1.500 | $6 d^{2} 7 s 7 p$ | ${ }^{5} D$ |  | 0 | 18127 | 18382 | 0.0 | 0.000 | 0.0 |
| $6 d^{3} 7 \mathrm{~s}$ | ${ }^{5} F$ | ${ }^{5} F$ | 2 | 6677 | 6362 | 1.010 | 1.001 | 1.000 | $6 d 7 s^{2} 7 p$ | ${ }^{3} P$ |  | 0 | 20790 | 20543 | 0.0 | 0.000 | 0.0 |
| $6 d^{2} 7 s^{2}$ |  | ${ }^{3} \mathrm{D}$ | 2 | 7445 | 7280 | 1.185 | 1.152 | 1.167 |  |  |  |  |  |  |  |  |  |
| $6 d^{3} 7 s$ | ${ }^{5} P$ | ${ }^{5} P$ | 2 | 12449 | 11802 | 1.780 | 1.759 | 1.833 | $6 d 7 s^{2} 7 p$ | ${ }^{3} D$ | ${ }^{3} D$ | 1 | 11455 | 11878 | 0.725 | 0.709 | 0.500 |
|  |  |  |  |  |  |  |  |  | $5 f 6 d 7 s^{2}$ | ${ }^{3} P$ | ${ }^{3} P$ | 1 | 13989 | 14244 | 1.205 | 1.218 | 1.500 |
| $6 d^{2} 7 s^{2}$ | ${ }^{3} P$ |  | 0 | 2708 | 2558 | 0.0 | 0.000 | 0.0 | $6 d^{2} 7 s 7 p$ | ${ }^{5} F$ | ${ }^{5} F$ | 1 | 15805 | 15737 | 0.385 | 0.297 | 0.000 |
| $6 d^{3} 7 s$ |  |  | 0 | 15243 | 14227 | 0.0 | 0.000 | 0.0 | $5 f 6 d 7 s^{2}$ | ${ }^{3} D$ | ${ }^{1} P$ | 1 | 17388 | 17357 | 0.505 | 1.024 | 1.000 |
| $6 d^{2} 7 s^{2}$ |  |  | 0 | 17327 | 16351 | 0.0 | 0.000 | 0.0 |  |  |  |  |  |  |  |  |  |
|  |  |  |  |  |  |  |  |  | $5 f 6 d 7 s^{2}$ | ${ }^{3} F$ | ${ }^{3} F$ | 2 | 7493 | 8244 | 0.775 | 0.796 | 0.667 |
| $6 d^{2} 7 s^{2}$ | ${ }^{3} P$ | ${ }^{3} P$ | 1 | 3948 | 3865 | 1.480 | 1.481 | 1.500 | $6 d 7 s^{2} 7 p$ | ${ }^{3} F$ | ${ }^{3} \mathrm{~F}$ | 2 | 10481 | 10783 | 0.725 | 0.730 | 0.667 |
| $6 d^{3} 7 s$ | ${ }^{5} F$ | ${ }^{5} F$ | 1 | 5887 | 5563 | 0.065 | 0.048 | 0.000 | $5 f 6 d 7 s^{2}$ | ${ }^{1} D$ | ${ }^{1} D$ | 2 | 11223 | 12114 | 0.975 | 0.921 | 1.000 |
| $6 d^{3} 7 s$ | ${ }^{5} P$ | ${ }^{5} P$ | 1 | 12179 | 11601 | 2.400 | 2.430 | 2.500 | $6 d 7 s^{2} 7 p$ | ${ }^{3} D$ | ${ }^{3} \mathrm{D}$ | 2 | 13641 | 14032 | 1.125 | 1.239 | 1.167 |
| $6 d^{3} 7 s$ |  | ${ }^{1} P$ | 1 | 14754 | 13963 | 0.760 | 0.714 | 1.000 | $6 d^{2} 7 s 7 p$ | ${ }^{5} G$ | ${ }^{3} \mathrm{~F}$ | 2 | 14253 | 14465 | 0.810 | 0.759 | 0.667 |
|  |  |  |  |  |  |  |  |  | $6 d 7 s^{2} 7 p$ | ${ }^{1} D$ | ${ }^{1} D$ | 2 | 15879 | 16217 | 1.070 | 1.085 | 1.000 |
| $6 d^{2} 7 s^{2}$ | ${ }^{3} F$ | ${ }^{3} F$ | 3 | 2815 | 2869 | 1.085 | 1.078 | 1.083 | $6 d^{2} 7 s 7 p$ | ${ }^{5} F$ | ${ }^{5} F$ | 2 | 17187 | 17224 | 1.045 | 1.029 | 1.000 |
| $6 d^{3} 7 \mathrm{~s}$ | ${ }^{5} F$ | ${ }^{5} F$ | 3 | 7818 | 7502 | 1.250 | 1.241 | 1.250 | $5 f 6 d 7 s^{2}$ | ${ }^{3} P$ | ${ }^{3} \mathrm{D}$ | 2 | 17781 | 17847 | 1.165 | 1.122 | 1.167 |
| $6 d^{3} 7 s$ | ${ }^{5} P$ | ${ }^{5} P$ | 3 | 13413 | 12848 | 1.390 | 1.629 | 1.667 |  |  |  |  |  |  |  |  |  |
| $6 d^{3} 7 s$ | ${ }^{3} G$ | ${ }^{3} G$ | 3 | 13613 | 13089 | 1.050 | 0.800 | 0.750 | $5 f 6 d 7 s^{2}$ | ${ }^{3} G$ | ${ }^{5} G$ | 3 | 10194 | 10527 | 0.870 | 0.940 | 0.917 |
| $6 d^{3} 7 s$ |  | ${ }^{3} F$ | 3 | 16685 | 15970 | 1.205 | 1.166 | 1.083 | $5 f 6 d 7 s^{2}$ | ${ }^{3} F$ | ${ }^{1} F$ | 3 | 10762 | 11242 | 1.010 | 0.914 | 1.000 |
| $6 d^{3} 7 s$ |  | ${ }^{5} F$ | 3 | 18138 | 17398 | 1.195 | 1.213 | 1.250 | $6 d 7 s^{2} 7 p$ | ${ }^{3} F$ | ${ }^{3} F$ | 3 | 13664 | 13945 | 1.110 | 1.129 | 1.083 |
| $6 d^{3} 7 s$ | ${ }^{3} F$ | ${ }^{3} F$ | 3 | 20527 | 19713 | 1.110 | 1.094 | 1.083 | $6 d^{2} 7 s 7 p$ | ${ }^{5} G$ | ${ }^{3} G$ | 3 | 15054 | 15167 | 1.065 | 0.591 | 0.750 |
| $6 d^{3} 7 s$ | ${ }^{1} F$ | ${ }^{5} F$ | 3 | 22222 | 21595 | 1.040 | 1.211 | 1.250 |  |  |  |  |  |  |  |  |  |
|  |  |  |  |  |  |  |  |  | $5 f 6 d 7 s^{2}$ | ${ }^{3} \mathrm{H}$ | ${ }^{3} \mathrm{H}$ | 4 | 7296 | 7795 | 0.865 | 0.872 | 0.800 |
| $6 d^{2} 7 s^{2}$ | ${ }^{3} F$ | ${ }^{3} F$ | 4 | 4953 | 4962 | 1.210 | 1.219 | 1.250 | $5 f 6 d 7 s^{2}$ | ${ }^{1} G$ | ${ }^{1} G$ | 4 | 9858 | 10414 | 0.985 | 0.962 | 1.000 |
| $6 d^{2} 7 s^{2}$ | ${ }^{1} G$ | ${ }^{5} G$ | 4 | 8156 | 8111 | 1.065 | 1.213 | 1.150 | $5 f 6 d 7 s^{2}$ | ${ }^{3} G$ | ${ }^{3} G$ | 4 | 12827 | 13175 | 1.095 | 1.111 | 1.050 |
| $6 d^{3} 7 s$ | ${ }^{5} F$ | ${ }^{5} G$ | 4 | 9113 | 8800 | 1.310 | 1.138 | 1.150 | $5 f 6 d 7 s^{2}$ | ${ }^{3} F$ | ${ }^{3} F$ | 4 | 13683 | 14207 | 1.170 | 1.135 | 1.250 |
| $6 d^{3} 7 s$ | ${ }^{3} G$ | ${ }^{1} G$ | 4 | 13782 | 13297 | 1.000 | 0.990 | 1.000 | $5 f 6 d^{2} 7 s$ | ${ }^{5} H$ | ${ }^{5} \mathrm{H}$ | 4 | 16092 | 16347 | 0.880 | 0.939 | 0.900 |
| $6 d^{3} 7 s$ | ${ }^{3} \mathrm{H}$ | ${ }^{3} \mathrm{H}$ | 4 | 15939 | 15493 | 0.905 | 0.887 | 0.800 | $5 f 6 d^{2} 7 s$ | ${ }^{5}$ | ${ }^{5}$ | 4 | 16840 | 16784 | 0.695 | 0.639 | 0.600 |
| $6 d^{3} 7 \mathrm{~s}$ | ${ }^{3} F$ | ${ }^{3} F$ | 4 | 18692 | 17960 | 1.175 | 1.209 | 1.250 | $6 d 7 s^{2} 7 p$ | ${ }^{3} F$ | ${ }^{3} F$ | 4 | 17883 | 18054 | 1.185 | 1.165 | 1.250 |
| $6 d^{3} 7 \mathrm{~s}$ | ${ }^{3} F$ | ${ }^{3} F$ | 4 | 20347 | 19532 | 1.204 | 1.196 | 1.250 | $6 d^{2} 7 s 7 p$ | ${ }^{5} G$ | ${ }^{5} G$ | 4 | 18503 | 18810 | 1.150 | 1.129 | 1.150 |
| $6 d^{3} 7 s$ |  | ${ }^{3} G$ | 4 | 21725 | 21646 | 1.090 | 1.056 | 1.050 |  |  |  |  |  |  |  |  |  |
|  |  |  |  |  |  |  |  |  | $5 f 6 d 7 s^{2}$ | ${ }^{3} \mathrm{H}$ | ${ }^{3} \mathrm{H}$ | 5 | 10884 | 11197 | 1.040 | 1.028 | 1.033 |
| $6 d^{3} 7 s$ | ${ }^{5} F$ | ${ }^{5} F$ | 5 | 10198 | 9805 | 1.365 | 1.360 | 1.400 | $5 f 6 d 7 s^{2}$ | ${ }^{3} G$ | ${ }^{3} G$ | 5 | 15255 | 15490 | 1.190 | 1.188 | 1.200 |
| $6 d^{3} 7 s$ |  | ${ }^{3} G$ | 5 | 14723 | 14204 | 1.150 | 1.134 | 1.200 | $5 f 6 d^{2} 7 s$ | ${ }^{5}$ I | ${ }^{5} \mathrm{H}$ | 5 | 17311 | 17501 | 1.015 | 1.099 | 1.100 |
| $6 d^{3} 7 s$ | ${ }^{3} \mathrm{H}$ | ${ }^{3} \mathrm{H}$ | 5 | 17614 | 17166 | 1.115 | 1.111 | 1.033 | $5 f 6 d^{2} 7 s$ | ${ }^{5} H$ | ${ }^{1} H$ | 5 | 17974 | 18011 | 1.025 | 0.942 | 1.000 |
| $6 d^{3} 7 s$ | ${ }^{1} H$ | ${ }^{1} H$ | 5 | 21775 | 21143 | 1.030 | 1.005 | 1.000 | $6 d^{2} 7 s 7 p$ | ${ }^{5} G$ | ${ }^{5} \mathrm{H}$ | 5 | 19320 | 19588 | 1.150 | 1.151 | 1.100 |
| $5 f 6 d 7 s 7 p$ | ${ }^{5} \mathrm{I}$ | ${ }^{1} H$ | 5 | 22845 | 23277 | 1.010 | 1.007 | 1.000 | $5 f 6 d^{2} 7 s$ | ${ }^{1} H$ | ${ }^{3} \mathrm{H}$ | 5 | 20586 | 20322 | 1.060 | 1.040 | 1.033 |
| $5 f 6 d 7 s 7 p$ | ${ }^{5} \mathrm{H}$ | ${ }^{3} \mathrm{H}$ | 5 | 26060 | 26381 | 1.025 | 1.031 | 1.033 | $6 d^{2} 7 s 7 p$ | ${ }^{5} \mathrm{~F}$ | ${ }^{5} G$ | 5 | 20932 | 21077 | 1.250 | 1.246 | 1.267 |
|  |  |  |  |  |  |  |  |  | $5 f 6 d^{2} 7 s$ | ${ }^{3} I$ | ${ }^{5}$ I | 5 | 22537 | 22399 | 0.930 | 0.893 | 0.900 |
| $6 d^{3} 7 s$ | ${ }^{3} \mathrm{H}$ | ${ }^{3} \mathrm{H}$ | 6 | 17063 | 16554 | 1.165 | 1.161 | 1.167 |  |  |  |  |  |  |  |  |  |
| $5 f 6 d 7 s 7 p$ | ${ }^{5} \mathrm{I}$ | ${ }^{5} \mathrm{I}$ | 6 | 26681 | 26997 | 1.110 | 1.109 | 1.071 | $5 f 6 d 7 s^{2}$ | ${ }^{3} \mathrm{H}$ | ${ }^{3} \mathrm{H}$ | 6 | 14188 | 14482 | 1.170 | 1.160 | 1.167 |
| $6 d^{4}$ | ${ }^{3} \mathrm{H}$ | ${ }^{3} \mathrm{H}$ | 6 | 28926 | 27972 | 1.125 | 1.164 | 1.167 | $5 f 6 d^{2} 7 s$ | ${ }^{5}$ I | ${ }^{5}$ I | 6 | 19332 | 19227 | 1.085 | 1.069 | 1.071 |
| $5 f 6 d 7 s 7 p$ | ${ }^{5} \mathrm{H}$ | ${ }^{3} \mathrm{H}$ | 6 | 29335 | 29553 | 1.185 | 1.139 | 1.167 | $5 f 6 d^{2} 7 s$ | ${ }^{5} \mathrm{H}$ | ${ }^{5} \mathrm{H}$ | 6 | 19940 | 19986 | 1.195 | 1.192 | 1.214 |
|  |  |  |  |  |  |  |  |  | $5 f 6 d^{2} 7 s$ | ${ }^{3} \mathrm{I}$ | ${ }^{3} \mathrm{I}$ | 6 | 23282 | 23307 |  | 1.029 | 1.024 |
|  |  |  |  |  |  |  |  |  | $6 d^{2} 7 s 7 p$ | ${ }^{5} G$ | ${ }^{5} G$ | 6 | 24038 | 24085 | 1.220 | 1.260 | 1.333 |
|  |  |  |  |  |  |  |  |  | $5 f 6 d 7 s^{2}$ | ${ }^{3} I$ | ${ }^{3} \mathrm{H}$ | 6 | 24861 | 24850 |  | 1.179 | 1.167 |

momentum, and $I$ the angular momentum of the nucleus. Thus $g_{F}=0$ when

$$
\begin{equation*}
F(F+1)=I(I+1)-J(J+1) \tag{7}
\end{equation*}
$$

subject to the triangularity condition between $I, J$, and $F$.

The simple expression of Eq. (7) belies the diversity of its solutions, as suggested in Table II. A number of the values of $I$ displayed there are associated with stable or long-lived nuclei.

## C. Energies of neutral Th

The calculated energy levels of thorium are in excellent agreement with experiment for such a heavy tetravalent neutral

TABLE II. Simplest nontrivial solutions of Eq. (7) for $F \leqslant 10$. The only solutions for $F<3 / 2$ are trivial. Multiple solutions are given for $F=4,5$, and 8 because they lie close together.

| $F$ | $J$ | $I$ | $F$ | $J$ | $I$ |
| :--- | :---: | :---: | :---: | :---: | :---: |
| $3 / 2$ | 3 | $7 / 2$ | 6 | 5 | 8 |
| 2 | 2 | 3 | $13 / 2$ | 8 | $21 / 2$ |
| $5 / 2$ | 8 | $17 / 2$ | 7 | $9 / 2$ | $17 / 2$ |
| 3 | 5 | 6 | $15 / 2$ | 11 | $27 / 2$ |
| $7 / 2$ | 4 | $11 / 2$ | 8 | $5 / 2$ | $17 / 2$ |
| 4 | $7 / 2$ | $11 / 2$ | 8 | $13 / 2$ | $21 / 2$ |
| 4 | 9 | 10 | 8 | 10 | 13 |
| $9 / 2$ | 7 | $17 / 2$ | $17 / 2$ | 80 | $171 / 2$ |
| 5 | 3 | 6 | 9 | 6 | 11 |
| 5 | 14 | 15 | $19 / 2$ | 12 | $31 / 2$ |
| $11 / 2$ | 35 | $71 / 2$ | 10 | 8 | 13 |

atom. We find $0.2 \%-2 \%$ differences between the theoretical and experimental energies for 35 out of 78 levels listed in Table I. Only 9 energies differ with experiment for more than $5 \%$. We find that inclusion of a sufficient number of configurations is particularly important for achieving accurate results. To ensure that all dominant configurations are included we started with the preliminary calculation that included a few thousand configuration state functions (CSFs). The initial configuration space was constructed by allowing two excitations into the valence orbitals from the $6 d^{2} 7 s^{2}, 6 d^{3} 7 s$, $6 d^{2} 7 s 7 p, 6 d 7 s^{2} 7 p$, and $5 f 6 d 7 s^{2}$ configurations. The results of that calculation allowed us to sort the configurations by their contributions to the energies of interest. Then we allowed two more excitations into the restricted valence space from the $\sim 35$ most important configurations and one more excitation into a large valence space from the $\sim 315$ most important configurations. The restricted valence space includes all orbitals with $l<5$ up to $10 g$ and the large valence space includes orbitals up to $20 d$. The combined file that includes all of the configurations of the initial run is constructed and duplicate entries are removed. The resulting list includes 24673 even and 28651 odd CSFs. This algorithm for construction of the configuration space was tested previously on tetravalent Hf [43], where the results of this approach were compared with results of much larger calculations.

The differences between the preliminary and final runs are small only for a first few even states, ranging from 0 to $300 \mathrm{~cm}^{-1}$. For most of the other states, the differences range between 900 and $1500 \mathrm{~cm}^{-1}$, with the average difference being $1250 \mathrm{~cm}^{-1}$. Moreover, the results of our preliminary runs show that lack of saturation of the CI space results in a $1500-2000 \mathrm{~cm}^{-1}$ shift of all $5 f 6 d 7 s^{2}$ and $5 f 6 d^{2} 7 s$ configurations relative to the even $7 s^{2} 6 d^{2}$ configuration. Selective expansion of the configuration space fixes this problem, resulting in the very good agrement between the energy levels of the $5 f 6 d 7 s^{2}$ and $5 f 6 d^{2} 7 s$ configurations with experiment.

The shift of the even vs odd configurations involving an $n f$ state is a well-known problem. For example, the energies of the $4 f^{6} 5 d$ levels in Gd IV [52] were shifted by $13500 \mathrm{~cm}^{-1}$
relative to the $4 f^{78} S_{7 / 2}$ ground level to account for this problem. The procedure for correcting the shift of the $4 f^{n} 5 d$ energies relative to the ground state $4 f^{n+1}$ energies was used for Nd IV, Pm IV, Sm IV, and Eu IV ions [53].

## D. Energies of Th ions

The energies and $g$ factors of 38 even-parity and 31 odd-parity states of trivalent $\mathrm{Th}^{+}$are listed in Table III. The table is structured in the same way as Table I. All values are counted from the $6 d^{2} 7 s^{2} D_{3 / 2}$ ground-state energy. We note that Ref. [23] does not assign the ground-state $L S J$ term designation. The corresponding experimental and our calculated ground-state $g$ factors listed in the first line of Table III, $g_{\text {expt }}=0.639$ and $g_{\text {present }}=0.662$, are in between the nonrelativistic values 0.4 and 0.8 for the ${ }^{4} F_{3 / 2}$ and ${ }^{2} D_{3 / 2}$ terms, respectively. We assign the ${ }^{2} D_{3 / 2}$ term designation to the ground state of $\mathrm{Th}^{+}$.

The 38 even-parity states listed in Table III belong to five configurations $6 d 7 s^{2}, 6 d^{2} 7 s, 6 d^{3}, 5 f^{2} 7 s$, and $5 f^{2} 6 d$, which are strongly mixed. Most odd-parity levels shown in Table III have a predominant composition of $5 f 6 d 7 s$. The CI plus all-order results are in good agreement with experiment for most states, with the differences being less than $5 \%$ for the $5 f 6 d 7 s$ states. The larger discrepancies with experiment that are observed for even-parity levels with two $5 f$ electrons, such as $5 f^{2} 7 s$ and $5 f^{2} 6 d$, are most likely due to the effects of correlations involving higher $(l>6)$ partial waves. This problem is exacerbated when two $5 f$ electrons are present in the same configuration.

Energies and $g$ factors for 95 levels of divalent $\mathrm{Th}^{2+}$ are listed in Table IV. The energies are counted from the ground state. The energies of the odd-parity $5 f 6 d, 5 f 7 d 5 f 7 s, 5 f 8 s$, $6 d 6 f$, and $6 d 7 p$ configurations and even-parity $6 d^{2}, 5 f^{2}, 7 s^{2}$, $5 f 7 p, 5 f 6 f$, and $6 d 7 s$ configurations calculated with the CI plus all-order approach are compared with experimental energies [23] given in columns "Present" and "Expt." of Table IV. We note that the ground state of $\mathrm{Th}^{2+}$ is $5 f 6 d^{3} H_{4}$ instead of the usual $n s^{2}{ }^{1} S_{0}$, such as in isoelectronic $\mathrm{Ac}^{+}$and Ra. The theoretical values agree well with experiment for most cases, with the exception of the $5 f 6 d^{3} F_{2}$ and $6 d^{2}{ }^{3} F_{2}$ levels that are very close to the ground state. Since we calculate these energies as the differences of the large ground and excited divalent removal energies, the accuracy is reduced for such small energy intervals. Our values agree with experiment to $0.10 \%-1 \%$ for 38 levels.

In Table V we compare all results for levels for which theoretical calculations were performed both by us and by Berengut et al. [33], where a CI plus MBPT approach was used. Berengut et al. [33] noted that "while we believe the $6 d^{2}{ }^{3} F_{3}$, $6 d^{2}{ }^{3} F_{4}$, and $6 d 7 s^{3} D_{3}$ transitions are accurate, the others are estimates only." Comparison of the two theoretical results with experimental data [23] given in Table V shows that our results for $\mathrm{Th}^{2+}$ are in better agreement with experiment than the CI plus MBPT values except for the $6 d^{2}{ }^{3} F_{3}$ and $6 d^{2}{ }^{3} F_{4}$ levels. The case of $\mathrm{Th}^{+}$is similar. The CI plus all-order method includes higher-order correlation beyond the CI plus MBPT approach. However, in some cases the higher-order terms may cancel with other contributions.

TABLE III. Levels (in $\mathrm{cm}^{-1}$ ) and $g$ factors of the lowest states of once-ionized thorium. Nonrelativistic values of $g$ factors $g_{\mathrm{nr}}$ are given by Eq. (3). Calculations are compared with experiment [12]. Configuration and term labels are determined as described in Sec. IV A.

|  | Even-parity states ( $\left.6 d^{2} 7 s+6 d^{3}+5 f^{2} 7 s+5 f^{2} 6 d\right)$ |  |  |  |  |  |  |  | Odd-parity states ( $\left.5 f 6 d 7 s+5 f 7 s^{2}+5 f 6 d^{2}\right)$ |  |  |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Conf. | Term |  |  | Energy |  | $g$ factor |  |  | Conf. | Term |  |  | Energy |  | $g$ factor |  |  |
|  | [23] | Present | $J$ | Present | Expt. | Expt. | Present | $g_{\text {nr }}$ |  | [23] | Present | $J$ | Present | Expt. | Expt. | Present | $g_{\text {nr }}$ |
| $6 d^{2} 7 s$ | ${ }^{4} F$ | ${ }^{2} D$ | 3/2 | 0 | 0 | 0.639 | 0.662 | 0.800 | $5 f 6 d 7 s$ | ${ }^{4} D$ | ${ }^{4} D$ | 1/2 | 11550 | 11725 | 0.255 | 0.239 | 0.000 |
| $6 d^{2} 7 s$ |  | ${ }^{4} F$ | 3/2 | 1948 | 1859 | 0.586 | 0.554 | 0.400 | $5 f 6 d 7 s$ |  | ${ }^{2} P$ | 1/2 | 13965 | 14102 | 0.523 | 0.552 | 0.667 |
| $6 d^{3}$ |  | ${ }^{2} D$ | 3/2 | 7779 | 7001 | 0.800 | 0.806 | 0.800 | $5 f 6 d 7 s$ | ${ }^{4} P$ | ${ }^{4} P$ | 1/2 | 15401 | 15324 | 2.565 | 2.567 | 2.667 |
| $6 d^{2} 7 s$ | ${ }^{4} P$ | ${ }^{4} P$ | 3/2 | 8622 | 8018 | 1.608 | 1.608 | 1.733 | $5 f 6 d 7 s$ |  | ${ }^{2} S$ | 1/2 | 17538 | 17838 | 1.08 | 1.582 | 2.000 |
| $6 d^{3}$ |  | ${ }^{2} D$ | 3/2 | 9209 | 8460 | 0.968 | 0.945 | 0.800 |  |  |  |  |  |  |  |  |  |
| $6 d^{2} 7 s$ | ${ }^{2} D$ | ${ }^{4} D$ | 3/2 | 13270 | 12220 | 0.977 | 0.946 | 1.200 | $5 f 6 d 7 s$ | ${ }^{4} F$ | ${ }^{4} F$ | 3/2 | 6020 | 6691 | 0.492 | 0.510 | 0.400 |
| $6 d^{3}$ | ${ }^{4} P$ | ${ }^{4} P$ | 3/2 | 16503 | 15237 | 1.592 | 1.612 | 1.733 | $5 f 6 d 7 s$ | ${ }^{2} D$ | ${ }^{2} D$ | 3/2 | 10695 | 11576 | 0.832 | 0.754 | 0.800 |
| $6 d^{3}$ |  | ${ }^{2} D$ | 3/2 | 19973 | 18119 | 0.93 | 0.862 | 0.800 | $5 f 6 d 7 s$ |  | ${ }^{4} D$${ }^{2} P$ | $3 / 2$ | 12657 | 12902 | 1.167 | 1.184 | 1.200 |
|  |  |  |  |  |  |  |  |  | $5 f 6 d 7 s$ |  |  | 3/2 | 15198 | 15145 | 1.366 | 1.278 | 1.333 |
| $6 d^{2} 7 s$ | ${ }^{4} P$ | ${ }^{4} P$ | 1/2 | 6972 | 6244 | 2.112 | 2.144 | 2.667 | $5 f 6 d 7 s$ | ${ }^{2} D$ | ${ }^{4} D$ | 3/2 | 15676 | 15711 | 1.06 | 1.162 | 1.200 |
| $6 d^{2} 7 s$ |  | ${ }^{2} S$ | 1/2 | 8509 | 7828 | 1.254 | 1.201 | 2.000 |  |  |  |  |  |  |  |  |  |
| $6 d^{3}$ | ${ }^{4} P$ | ${ }^{4} P$ | 1/2 | 15632 | 14349 | 2.555 | 2.564 | 2.667 | $5 f 7 s^{2}$ | ${ }^{2} F$ | ${ }^{2} F$ | 5/2 | 3882 | 4490 | 0.856 | 0.853 | 0.857 |
|  |  |  |  |  |  |  |  |  | $5 f 6 d 7 s$ | ${ }^{4} F$ | ${ }^{4} F$ | 5/2 | 6650 | 7331 | 1.061 | 1.070 | 1.029 |
| $6 d^{2} 7 s$ | ${ }^{4} F$ | ${ }^{4} F$ | 5/2 | 1722 | 1522 | 1.076 | 1.070 | 1.029 | $5 f 6 d 7 s$ | ${ }^{4} G$ | ${ }^{4} G$ | 5/2 | 9299 | 9585 | 0.601 | 0.606 | 0.571 |
| $6 d 7 s^{2}$ |  | ${ }^{2} D$ | 5/2 | 4185 | 4113 | 1.163 | 1.150 | 1.200 | $5 f 6 d 7 s$ |  | ${ }^{4} F$ | 5/2 | 9978 | 10673 | 1.088 | 1.089 | 1.029 |
| $6 d^{2} 7 s$ | ${ }^{2} F$ | ${ }^{2} F$ | 5/2 | 9198 | 8606 | 0.986 | 0.982 | 0.857 | $5 f 6 d 7 s$ | ${ }^{2} F$ | ${ }^{4} F$ | 5/2 | 12045 | 12472 | 0.982 | 0.935 | 1.029 |
| $6 d^{2} 7 s$ | ${ }^{4} P$ | ${ }^{4} P$ | 5/2 | 9680 | 9061 | 1.419 | 1.408 | 1.600 | $5 f 6 d 7 s$ | ${ }^{4} D$ | ${ }^{4} D$ | 5/2 | 14310 | 14546 | 1.339 | 1.346 | 1.371 |
| $6 d^{3}$ |  | ${ }^{4} F$ | 5/2 | 10145 | 9401 | 1.034 | 1.035 | 1.029 |  |  |  |  |  |  |  |  |  |
| $6 d^{2} 7 s$ |  | ${ }^{2} D$ | 5/2 | 14220 | 13251 | 1.245 | 1.235 | 1.200 | $5 f 6 d 7 s$ | ${ }^{4} \mathrm{H}$ | ${ }^{4} \mathrm{H}$ | 7/2 | 5743 | 6168 | 0.718 | 0.729 | 0.667 |
| $6 d^{3}$ | ${ }^{4} P$ | ${ }^{4} P$ | 5/2 | 17100 | 15787 | 1.571 | 1.566 | 1.600 | $5 f 7 s^{2}$ | ${ }^{2} F$ | ${ }^{2} F$ | 7/2 | 7794 | 8379 | 1.132 | 1.127 | 1.143 |
| $6 d^{3}$ | ${ }^{2} D$ | ${ }^{4} D$ | 5/2 | 22144 | 20159 | 1.19 | 1.189 | 1.371 | $5 f 6 d 7 s$ | ${ }^{2} G$ | ${ }^{2} G$ | 7/2 | 8681 | 9202 | 0.911 | 0.899 | 0.889 |
|  |  |  |  |  |  |  |  |  | $5 f 6 d 7 s$ | ${ }^{4} F$ | ${ }^{4} F$ | 7/2 | 9304 | 9720 | 1.173 | 1.167 | 1.238 |
| $6 d^{2} 7 s$ | ${ }^{4} F$ | ${ }^{4} F$ | 7/2 | 4374 | 4147 | 1.232 | 1.227 | 1.238 | $5 f 6 d 7 s$ | ${ }^{4} G$ | ${ }^{4} G$ | 7/2 | 10751 | 11117 | 0.983 | 0.977 | 0.984 |
| $6 d^{2} 7 s$ | ${ }^{2} G$ | ${ }^{4} G$ | 7/2 | 10502 | 9712 | 0.953 | 0.947 | 0.984 | $5 f 6 d^{2}$ |  | ${ }^{2} F$ | 7/2 | 13270 | 12486 | 0.855 | 1.036 | 1.143 |
| $6 d^{3}$ |  | ${ }^{2} F$ | 7/2 | 11621 | 10855 | 1.166 | 1.171 | 1.143 |  |  |  |  |  |  |  |  |  |
| $6 d^{3}$ |  | ${ }^{2} F$ | 7/2 | 13297 | 12570 | 1.131 | 1.122 | 1.143 | $5 f 6 d 7 s$ |  | ${ }^{4} \mathrm{H}$ | 9/2 | 6265 | 6700 | 1.018 | 1.025 | 0.970 |
| $6 d^{3}$ | ${ }^{2} G$ | ${ }^{2} G$ | 7/2 | 18247 | 16818 | 0.916 | 0.906 | 0.889 | $5 f 6 d 7 s$ |  | ${ }^{2} G$ | 9/2 | 8818 | 9238 | 1.086 | 1.068 | 1.111 |
| $6 d^{3}$ | ${ }^{2} F$ | ${ }^{2} F$ | 7/2 | 25165 | 22834 | 1.12 | 1.132 | 1.143 | $5 f 6 d 7 s$ |  | ${ }^{4} \mathrm{H}$ | 9/2 | 10435 | 10572 | 0.931 | 0.927 | 0.970 |
|  |  |  |  |  |  |  |  |  | $5 f 6 d 7 s$ |  | ${ }^{4} F$ | 9/2 | 12076 | 12488 | 1.245 | 1.253 | 1.333 |
| $6 d^{2} 7 s$ | ${ }^{4} F$ | ${ }^{4} F$ | 9/2 | 6528 | 6213 | 1.312 | 1.309 | 1.333 |  | ${ }^{4} G$ | ${ }^{4} G$ | 9/2 | 13156 | 13469 | 1.185 | 1.178 | 1.172 |
| $6 d^{2} 7 s$ |  | ${ }^{2} G$ | 9/2 | 11158 | 10379 | 1.153 | 1.145 | 1.111 | $5 f 6 d^{2}$ | ${ }^{4} \mathrm{H}$ | ${ }^{2} H$ | 9/2 | 14421 | 15243 | 1.00 | 0.839 | 0.909 |
| $6 d^{3}$ | ${ }^{4} F$ | ${ }^{4} F$ | 9/2 | 14217 | 13249 | 1.242 | 1.256 | 1.333 |  |  |  |  |  |  |  |  |  |
| $6 d^{3}$ | ${ }^{2} H$ | ${ }^{4} \mathrm{H}$ | 9/2 | 16585 | 15305 | 1.006 | 0.983 | 0.970 | $5 f 6 d 7 s$ | ${ }^{4} H$ | ${ }^{4} H$ | 11/2 | 9953 | 10189 | 1.128 | 1.121 | 1.133 |
| $6 d^{3}$ | ${ }^{2} G$ | ${ }^{2} G$ | 9/2 | 21328 | 19880 | 1.08 | 1.075 | 1.111 | $5 f 6 d 7 s$ | ${ }^{4} G$ | ${ }^{2} H$ | 11/2 | 14390 | 15350 | 1.267 | 1.085 | 1.091 |
| $5 f^{2} 7 s$ |  | ${ }^{2} H$ | 9/2 | 30481 | 25246 | 0.96 | 0.939 | 0.909 |  |  |  | 11/2 | 15111 | 16565 | 0.98 | 1.261 | 1.273 |
|  |  |  |  |  |  |  |  |  | $5 f 6 d^{2}$ | ${ }^{4} \mathrm{H}$ | ${ }^{4} I$ | 11/2 | 16717 | 17771 | 1.10 | 0.966 | 0.965 |
| $6 d^{3}$ | ${ }^{2} \mathrm{H}$ | ${ }^{4} H$ | 11/2 | 19039 | 17727 | 1.09 | 1.086 | 1.091 |  |  |  |  |  |  |  |  |  |
| $5 f^{2} 7 s$ | ${ }^{4} \mathrm{H}$ | ${ }^{2} \mathrm{H}$ | 11/2 | 33216 | 27937 | 1.12 | 1.118 | 1.133 |  |  |  |  |  |  |  |  |  |
| $5 f^{2} 7 s$ | ${ }^{2} H$ | ${ }^{4} \mathrm{H}$ | 11/2 | 35702 | 30485 | 1.08 | 1.085 | 1.091 |  |  |  |  |  |  |  |  |  |
| $5 f^{2} 7 s$ | ${ }^{4} \mathrm{~K}$ | ${ }^{2} I$ | 11/2 | 37981 | 32621 | 0.826 | 0.885 | 0.923 |  |  |  |  |  |  |  |  |  |
| $5 f^{2} 7 s$ | ${ }^{4} H$ | ${ }^{4} \mathrm{H}$ | 13/2 | 35827 | 30549 | 1.23 | 1.221 | 1.231 |  |  |  |  |  |  |  |  |  |
| $5 f^{2} 6 d$ | ${ }^{4} K$ | ${ }^{4} K$ | 13/2 | 41108 | 35401 | 0.98 | 0.973 | 0.964 |  |  |  |  |  |  |  |  |  |
| $5 f^{2} 7 s$ | ${ }^{2} I$ | ${ }^{4} I$ | 13/2 | 42291 | 37575 | 1.088 | 1.120 | 1.108 |  |  |  |  |  |  |  |  |  |

The CI plus MBPT approach was used by Porsev and Flambaum [31] to evaluate energies and $g$ factors in $\mathrm{Th}^{+}$. Tabulated results were given for even-parity states with $J=3 / 2$ and $5 / 2$ in the range from 18119 to $40644 \mathrm{~cm}^{-1}$. Since our calculation was carried out for lower levels, we can only compare results for two states $6 d^{32} D_{3 / 2}$ and $6 d^{34} D_{5 / 2}$. The CI plus MBPT results of [31] differ from the experimental values by $18 \%$ and our CI plus all-order values by $10 \%$.

## V. MULTIPOLE TRANSITION AMPLITUDES AND LIFETIMES IN Ra-LIKE Th ${ }^{2+}$

We now discuss some multipole transition amplitudes in $\mathrm{Th}^{2+}$ that are representative of the calculations that may have to be done for the investigation of the electronic bridge process [11]. Our CI plus all-order results for the multipole matrix elements, transition rates, and lifetimes in Ra-like $\mathrm{Th}^{2+}$ are given in Table VI.

TABLE IV. Levels (in $\mathrm{cm}^{-1}$ ) and $g$ factors of the lowest states of twice-ionized thorium. Nonrelativistic values of $g$ factors $g_{\mathrm{nr}}$ are given by Eq. (3). Calculations are compared with experiment [12]. Configuration and term labels are determined as described in Sec. IV A.

| Odd-parity states ( $5 f 6 d+5 f 7 d+5 f 7 s+5 f 8 s+6 d 6 f+6 d 7 p)$ |  |  |  |  |  |  |  |  | Even-parity states ( $\left.6 d^{2}+5 f^{2}+7 s^{2}+5 f 7 p+5 f 6 f+6 d 7 s\right)$ |  |  |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Conf. | Term |  |  | Energy |  | $g$ factors |  |  | Conf. | Term |  |  | Energy |  | $g$ factors |  |  |
|  | [23] | Present | $J$ | Present | Expt. | Expt. | Present | $g_{\mathrm{nr}}$ |  | [23] | Present | $J$ | Present | Expt. | Expt. | Present | $g_{\mathrm{nr}}$ |
| $5 f 6 d$ | ${ }^{3} \mathrm{H}$ | ${ }^{3} \mathrm{H}$ | 4 | 0 | 0.0 | 0.885 | 0.894 | 0.800 | $6 d^{2}$ | ${ }^{3} P$ |  | 0 | 6151 | 5090 | 0.00 | 0.000 | 0.000 |
| $5 f 6 d$ | ${ }^{3} F$ | ${ }^{1} G$ | 4 | 3207 | 3188 | 0.976 | 0.949 | 1.000 | $7 s^{2}$ | ${ }^{1} S$ |  | 0 | 12428 | 11961 | 0.00 | 0.000 | 0.000 |
| $5 f 7 s$ | ${ }^{3} F$ | ${ }^{3} F$ | 4 | 6237 | 6311 | 1.22 | 1.205 | 1.250 | $6 d^{2}$ | ${ }^{1} S$ |  | 0 | 22008 | 18993 | 0.00 | 0.000 | 0.000 |
| $5 f 6 d$ | ${ }^{3} G$ | ${ }^{3} G$ | 4 | 8197 | 8142 | 1.103 | 1.096 | 1.050 | $5 f^{2}$ | ${ }^{3} P$ |  | 0 | 29579 | 29299 | 0.00 | 0.000 | 0.000 |
| $5 f 6 d$ | ${ }^{3} \mathrm{H}$ | ${ }^{3} F$ | 4 | 9063 | 8981 | 1.188 | 1.178 | 1.250 | $5 f^{2}$ | ${ }^{1} S$ |  | 0 | 55356 | 51161 | 0.00 | 0.000 | 0.000 |
| $6 d 7 p$ |  | ${ }^{3} F$ | 4 | 54333 | 53052 | 1.27 | 1.243 | 1.250 |  |  |  | 0 | 84509 |  | 0.00 | 0.000 | 0.000 |
| $5 f 7 d$ |  | ${ }^{3} \mathrm{H}$ | 4 | 79062 | 78417 |  | 0.891 | 0.800 | $5 f 6 f$ |  |  | 0 | 89234 | 88313 | 0.00 | 0.000 | 0.000 |
| $5 f 8 s$ |  | ${ }^{3} F$ | 4 | 79519 | 78930 |  | 1.210 | 1.250 |  |  |  | 0 | 94112 |  | 0.00 | 0.000 | 0.000 |
| $5 f 6 d$ | ${ }^{3} P$ |  | 0 | 11766 | 11233 | 0.0 | 0.000 | 0.000 | $6 d 7 s$ | ${ }^{3} D$ | ${ }^{3} D$ | 1 | 6137 | 5524 | 0.50 | 0.499 | 0.500 |
| $7 s 7 p$ | ${ }^{3} P$ |  | 0 | 43188 | 42260 | 0.0 | 0.000 | 0.000 | $6 d^{2}$ | ${ }^{3} P$ | ${ }^{3} P$ | 1 | 8905 | 7876 | 1.50 | 1.491 | 1.500 |
| $6 d 7 p$ |  |  | 0 | 52776 | 51745 | 0.0 | 0.000 | 0.000 | $5 f^{2}$ | ${ }^{3} P$ | ${ }^{3} P$ | 1 | 30636 | 30402 | 1.494 | 1.492 | 1.500 |
| $5 f 7 d$ |  |  | 0 | 81644 | 80906 | 0.0 | 0.000 | 0.000 | $5 f 7 p$ |  | ${ }^{3} D$ | 1 | 44946 | 44603 | 0.495 | 0.497 | 0.500 |
| $5 f 6 d$ | ${ }^{3} D$ | ${ }^{3} D$ | 1 | 8260 | 7921 | 0.621 | 0.599 | 0.500 | $6 d^{2}$ | ${ }^{3} F$ | ${ }^{3} F$ | 2 | 895 | 63 | 0.744 | 0.756 | 0.667 |
| $5 f 6 d$ | ${ }^{3} P$ | ${ }^{3} P$ | 1 | 11564 | 11123 | 1.352 | 1.376 | 1.500 | $6 d^{2}$ |  | ${ }^{1} D$ | 2 | 5426 | 4676 | 1.020 | 1.003 | 1.000 |
| $5 f 6 d$ | ${ }^{1} P$ | ${ }^{1} P$ | 1 | 22733 | 20711 |  | 1.006 | 1.000 | $6 d 7 s$ | ${ }^{3} \mathrm{D}$ | ${ }^{3} D$ | 2 | 7943 | 7176 | 1.180 | 1.155 | 1.167 |
| $6 d 7 p$ |  | ${ }^{1} P$ | 1 | 40282 | 39281 | 0.911 | 0.902 | 1.000 | $6 d^{2}$ | ${ }^{3} P$ | ${ }^{3} P$ | 2 | 11417 | 10440 | 1.36 | 1.370 | 1.500 |
| $7 s 7 p$ | ${ }^{3} P$ | ${ }^{3} P$ | 1 | 45938 | 45064 | 1.120 | 1.109 | 1.500 | $6 d 7 s$ | ${ }^{1} D$ | ${ }^{1} D$ | 2 | 16438 | 16037 | 1.00 | 0.961 | 1.000 |
| $6 d 7 p$ |  | ${ }^{3} P$ | 1 | 51875 | 50993 | 1.22 | 1.202 | 1.500 | $5 f^{2}$ | ${ }^{3} \mathrm{~F}$ | ${ }^{3} F$ | 2 | 18616 | 18864 | 0.694 | 0.738 | 0.667 |
| $6 d 7 p$ |  | ${ }^{3} P$ | 1 | 55035 | 53939 | 1.25 | 1.248 | 1.500 | $5 f^{2}$ | ${ }^{1} D$ | ${ }^{3} \mathrm{D}$ | 2 | 28971 | 28233 | 1.12 | 1.164 | 1.167 |
| $7 s 7 p$ | ${ }^{1} P$ | ${ }^{1} P$ | 1 | 69930 | 69001 |  | 1.012 | 1.000 | $5 f^{2}$ | ${ }^{3} P$ | ${ }^{3} P$ | 2 | 33488 | 32867 | 1.344 | 1.300 | 1.500 |
| $5 f 6 d$ | ${ }^{3} F$ | ${ }^{3} F$ | 2 | 189 | 511 | 0.711 | 0.765 | 0.667 | $6 d^{2}$ | ${ }^{3} F$ | ${ }^{3} F$ | 3 | 4938 | 4056 | 1.083 | 1.078 | 1.083 |
| $5 f 6 d$ | ${ }^{3} F$ | ${ }^{3} F$ | 2 | 2958 | 3181 | 0.725 | 0.732 | 0.667 | $6 d 7 s$ | ${ }^{3} \mathrm{D}$ | ${ }^{3} \mathrm{D}$ | 3 | 10641 | 9954 | 1.339 | 1.325 | 1.333 |
| $5 f 6 d$ | $J$ | ${ }^{1} D$ | 2 | 5797 | 6288 | 0.908 | 0.858 | 1.000 | $5 f^{2}$ | ${ }^{3} F$ | ${ }^{3} F$ | 3 | 20378 | 20840 | 1.096 | 1.078 | 1.083 |
| $5 f 6 d$ | ${ }^{3} D$ | ${ }^{3} D$ | 2 | 10458 | 10181 | 1.19 | 1.181 | 1.167 | $5 f 7 p$ |  | ${ }^{3} G$ | 3 | 33715 | 33562 | 0.849 | 0.837 | 0.750 |
| $5 f 6 d$ | ${ }^{3} P$ | ${ }^{3} P$ | 2 | 13513 | 13208 | 1.432 | 1.436 | 1.500 | $5 f 7 p$ |  | ${ }^{3} F$ | 3 | 38736 | 38432 | 1.170 | 1.149 | 1.083 |
| $6 d 7 p$ |  | ${ }^{3} F$ | 2 | 38322 | 37280 | 0.795 | 0.782 | 0.667 | $5 f 7 p$ |  | ${ }^{1} F$ | 3 | 42544 | 42313 | 0.971 | 0.960 | 1.000 |
| $6 d 7 p$ |  | ${ }^{3} D$ | 2 | 45159 | 44088 | 1.200 | 1.187 | 1.167 | $5 f 7 p$ |  | ${ }^{3} F$ | 3 | 47876 | 47472 | 1.202 | 1.199 | 1.083 |
| $6 d 7 p$ |  | ${ }^{1} D$ | 2 | 48723 | 47680 | 1.02 | 1.002 | 1.000 | $6 d 7 d$ |  | ${ }^{1} F$ | 3 | 84838 | 83702 |  | 0.906 | 1.000 |
| $5 f 7 s$ |  | ${ }^{3} F$ | 3 | 2436 | 2527 | 1.071 | 1.052 | 1.083 | $6 d^{2}$ | ${ }^{3} F$ | ${ }^{3} F$ | 4 | 7264 | 6538 | 1.20 | 1.161 | 1.250 |
| $5 f 6 d$ |  | ${ }^{1} F$ | 3 | 4853 | 4827 | 1.003 | 0.988 | 1.000 | $6 d^{2}$ | ${ }^{1} G$ | ${ }^{3} G$ | 4 | 10822 | 10543 | 1.05 | 1.064 | 1.050 |
| $5 f 6 d$ | ${ }^{3} G$ | ${ }^{3} G$ | 3 | 5085 | 5061 | 0.869 | 0.858 | 0.750 | $5 f^{2}$ | ${ }^{3} \mathrm{H}$ | ${ }^{3} \mathrm{H}$ | 4 | 14514 | 15149 | 0.81 | 0.814 | 0.800 |
| $5 f 7 s$ |  | ${ }^{1} F$ | 3 | 7609 | 7501 | 1.027 | 1.015 | 1.000 | $5 f^{2}$ | ${ }^{3} \mathrm{~F}$ | ${ }^{3} F$ | 4 | 21782 | 21784 | 1.18 | 1.201 | 1.250 |
| $5 f 6 d$ | ${ }^{3} D$ | ${ }^{3} D$ | 3 | 11236 | 10741 | 1.22 | 1.244 | 1.333 | $5 d 2$ | ${ }^{1} G$ | ${ }^{1} G$ | 4 | 27045 | 25972 | 1.072 | 1.033 | 1.000 |
| $5 f 6 d$ | ${ }^{1} F$ | ${ }^{3} F$ | 3 | 16506 | 15453 | 1.07 | 1.060 | 1.083 | $5 f 7 p$ |  | ${ }^{3} G$ | 4 | 38980 | 38581 | 1.105 | 1.100 | 1.050 |
| $6 d 7 p$ |  | ${ }^{3} F$ | 3 | 45686 | 44465 | 1.125 | 1.121 | 1.083 | $5 f 7 p$ |  | ${ }^{3} G$ | 4 | 44034 | 43702 | 1.069 | 1.068 | 1.050 |
| $6 d 7 p$ |  | ${ }^{3} D$ | 3 | 51059 | 49981 | 1.19 | 1.188 | 1.333 | $5 f 7 p$ |  | ${ }^{3} F$ | 4 | 47745 | 47261 | 1.14 | 1.114 | 1.250 |
| $5 f 6 d$ | ${ }^{3} \mathrm{H}$ | ${ }^{3} \mathrm{H}$ | 5 | 4802 | 4490 | 1.04 | 1.028 | 1.033 | $5 f^{2}$ | ${ }^{3} \mathrm{H}$ | ${ }^{3} \mathrm{H}$ | 5 | 17131 | 17888 | 1.01 | 1.028 | 1.033 |
| $5 f 6 d$ | ${ }^{3} G$ | ${ }^{3} G$ | 5 | 11456 | 11277 | 1.186 | 1.187 | 1.200 | $5 f 7 p$ |  | ${ }^{3} G$ | 5 | 47781 | 47422 | 1.207 | 1.194 | 1.200 |
| $5 f 6 d$ | ${ }^{1} H$ | ${ }^{1} H$ | 5 | 20144 | 19009 | 1.001 | 1.001 | 1.000 | $5 f 6 f$ |  | ${ }^{3} I$ | 5 | 87443 | 86934 |  | 0.887 | 0.833 |
| $5 f 7 d$ |  | ${ }^{1} H$ | 5 | 80841 | 80137 |  | 1.033 | 1.000 | $5 f 6 f$ |  | ${ }^{1} H$ | 5 | 88176 | 87667 |  | 1.006 | 1.000 |
| $5 f 7 d$ |  | ${ }^{3} G$ | 5 | 83847 | 83023 |  | 1.137 | 1.200 | $6 d 7 d$ |  | ${ }^{3} G$ | 5 | 90979 | 90085 |  | 1.141 | 1.200 |
| $5 f 7 d$ |  | ${ }^{3} \mathrm{H}$ | 5 | 85158 | 84239 |  | 1.047 | 1.033 | $5 f 6 f$ |  | ${ }^{3} \mathrm{H}$ | 5 | 92840 | 92103 |  | 1.020 | 1.033 |
| $6 d 6 f$ |  | ${ }^{3} \mathrm{H}$ | 5 | 97802 | 96317 |  | 1.049 | 1.033 | $5 f 8 p$ |  | ${ }^{3} G$ | 5 | 94046 | 94144 |  | 1.186 | 1.200 |
| $5 f 6 d$ | ${ }^{3} \mathrm{H}$ | ${ }^{3} \mathrm{H}$ | 6 | 8810 | 8437 | 1.17 | 1.160 | 1.167 | $5 f^{2}$ | ${ }^{3} \mathrm{H}$ | ${ }^{3} \mathrm{H}$ | 6 | 20123 | 20771 | 1.16 | 1.157 | 1.167 |
| $5 f 7 d$ |  | ${ }^{3} \mathrm{H}$ | 6 | 84721 | 83963 |  | 1.161 | 1.167 | $5 f^{2}$ | ${ }^{1}$ I | ${ }^{1}$ I | 6 | 28635 | 28350 |  | 0.999 | 1.000 |
|  |  | ${ }^{3} \mathrm{H}$ | 6 | 102352 |  |  | 1.160 | 1.167 | $5 f 6 f$ |  | ${ }^{3} \mathrm{I}$ | 6 | 88845 | 88387 |  | 1.021 | 1.024 |
|  |  | ${ }^{3}$ I | 6 | 108394 |  |  | 1.032 | 1.024 | $5 f 6 f$ |  | ${ }^{3} \mathrm{H}$ | 6 | 93648 | 93045 |  | 1.101 | 1.167 |
|  |  | ${ }^{3} \mathrm{~K}$ | 6 | 108614 |  |  | 0.868 | 0.857 | $5 f 6 f$ |  | ${ }^{3} \mathrm{I}$ | 6 | 94584 | 94018 |  | 1.052 | 1.024 |

TABLE V. Levels (in $\mathrm{cm}^{-1}$ ) of the lowest states of once- and twice-ionized thorium. All energy values are given relative to the respective ground state. Calculations are compared with experiment [12] and other theory [33].

| $\mathrm{Th}^{+}$ | Term | Energies |  |  | Th ${ }^{2+}$ | Term | Energies |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  | Present | Expt. [12] | Theory [33] |  |  | Present | Expt. [12] | Theory [33] |
| $5 f 7 s^{2}$ | ${ }^{2} F_{5 / 2}$ | 3882 | 4490 | 4856 | $6 d^{2}$ | ${ }^{3} F_{3}$ | 4938 | 4056 | 4023 |
| $5 f 6 d 7 s$ | ${ }^{4} F_{3 / 2}$ | 6020 | 6691 | 7487 | $6 d^{2}$ | ${ }^{3} F_{4}$ | 7264 | 6538 | 6795 |
| $5 f 6 d 7 s$ | ${ }^{4} F_{5 / 2}$ | 6651 | 7331 | 8325 | $6 d 7 s$ | ${ }^{3} D_{3}$ | 10641 | 9954 | 9204 |
| $5 f 6 d 7 s$ | ${ }^{4} G_{5 / 2}$ | 9229 | 9585 | 10045 | $6 d^{2}$ | ${ }^{1} G_{4}$ | 10822 | 10543 | 11051 |
| $5 f 6 d 7 s$ | ${ }^{4} H_{5 / 2}$ | 9978 | 10673 | 12168 | $5 f^{2}$ | ${ }^{3} H_{4}$ | 14514 | 15149 | 13358 |
| $5 f 6 d 7 s$ | ${ }^{2} D_{3 / 2}$ | 10695 | 11576 | 13054 | $5 f^{2}$ | ${ }^{3} H_{5}$ | 17131 | 17887 | 16068 |
| $5 f 6 d 7 s$ | ${ }^{4} D_{1 / 2}$ | 11575 | 11725 | 12897 | $5 f^{2}$ | ${ }^{3} F_{3}$ | 20378 | 20840 | 19080 |
| $5 f 6 d 7 s$ | ${ }^{2} F_{5 / 2}$ | 12045 | 12472 | 14564 | $5 f^{2}$ | ${ }^{3} F_{4}$ | 21782 | 21784 | 20366 |
| $5 f 6 d 7 s$ | ${ }^{4} F_{3 / 2}$ | 12657 | 12902 | 14233 | $5 f^{2}$ | ${ }^{1} G_{4}$ | 27045 | 25972 | 25269 |
| $5 f 6 d 7 s$ | ${ }^{4} G_{1 / 2}$ | 13965 | 14102 | 15853 | $5 f 7 p$ | ${ }^{3} G_{3}$ | 33715 | 33562 | 33402 |
|  |  |  |  |  | $5 f 7 p$ | ${ }^{3} F_{3}$ | 38736 | 38432 | 38617 |

We evaluate multipole matrix elements between 12 oddparity states with energies in the $0-8142 \mathrm{~cm}^{-1}$ range with 12 even-parity states with energies in the 63-11961 cm ${ }^{-1}$ range. This results in $45 E 1,66 M 2$, and $82 E 3$ transitions between the odd-even and even-odd states. We evaluate also the $83 M 1$, $E 2$, and $M 3$ transitions inside the even-parity complex, as well as the $110 M 1, E 2$, and $M 3$ transitions inside the odd-parity complex. That gives us 386 multipole matrix elements for transitions between lowest-lying levels in the $\mathrm{Th}^{2+}$ ion.

In Table VI we include results for 45 selected electric multipole ( $E 1$ and $E 2$ ) and magnetic multipole ( $M 1$ and $M 2$ ) transitions that are most important for the evaluation of the corresponding lifetimes. The octupole ( $E 3$ and $M 3$ ) transitions make negligible contributions to the lifetimes and are omitted.

We use atomic units to express all transition matrix elements throughout this section: The numerical values of the elementary charge $e$, the reduced Planck constant $\hbar=h / 2 \pi$, and the electron mass $m_{e}$ are set equal to 1 . The atomic unit for the electric dipole matrix element is $e a_{0}$, where $a_{0}$ is the Bohr radius.

To show the importance of using effective transition operators (for example, electric dipole $D^{\text {eff }}$ ), which include random-phase-approximation (RPA) corrections, instead of the bare operators we give the matrix elements with and without the RPA correction in columns labeled $Z^{\mathrm{CI}+\text { all-order }}$ and $Z^{\text {no RPA }}$, respectively. We find that the RPA correction is significant ( $20 \%-50 \%$ ) for most transitions, so the $Z^{\text {CI+all-order }}$ final values are used in calculating transition rates and lifetimes. The RPA correction is small for $M 1$ matrix elements $5 f 7 s^{3} F_{3}-5 f 7 s^{3} F_{2}$ and $6 d^{2}{ }^{3} P_{1}-7 s^{2}{ }^{1} S_{0}$.

The $E 1, E 2, E 3, M 2, M 3$, and $M 3$ transition probabilities $A_{r}\left(\mathrm{~s}^{-1}\right)$ are obtained in terms of line strengths $S$ (a.u.) and energies $\mathcal{E}$ (a.u.) as

$$
\begin{aligned}
A(E k) & =\frac{C^{(k)}[\mathcal{E}]^{2 k+1}}{(2 J+1)} S(E k), \\
C^{(1)} & =2.14200 \times 10^{10} \\
C^{(2)} & =5.70322 \times 10^{4} \\
C^{(3)} & =7.71311 \times 10^{-2}
\end{aligned}
$$

$$
\begin{align*}
A(M k) & =\frac{D^{(k)}[\mathcal{E}]^{2 k+1}}{(2 J+1)} S(M k), \\
D^{(1)} & =2.85161 \times 10^{5} \\
D^{(2)} & =7.59260 \times 10^{-1}  \tag{9}\\
D^{(3)} & =1.02683 \times 10^{-6}
\end{align*}
$$

The line strengths $S(E 1), S(E 2), S(E 1), S(E 3), S(M 1)$, $S(M 2)$, and $S(M 3)$ are obtained as squares of the corresponding $E 1, E 2, E 3, M 2, M 3$, and $M 3$ matrix elements listed in column $Z^{\mathrm{CI}+\text { all }}$ of Table VI. Energies are from the experimental compilation of Ref. [23]. We list the experimental energies for upper and lower states as well as the corresponding transition wavelengths $\lambda$ in Table VI for reference. Our results for the transition rates are given in column $A_{r}^{\mathrm{C}+\text { all }}$ of Table VI.

In order to determine the lifetimes listed in the last column of Table VI, we sum over all possible radiative transitions. The number of contributing transitions increases significantly for higher levels. For example, 18 transitions contribute to the lifetime of the relatively low-lying $5 f 7 s^{3} F_{4}$ state $E\left(5 f 7 s^{3} F_{4}\right)=$ $6310.81 \mathrm{~cm}^{-1}$. However, only one transition $6 d^{2}{ }^{3} F_{3}-5 f 7 s^{3} F_{4}$ contributes significantly and the total contribution of other 17 transitions to the $5 f 7 s^{3} F_{4}$ lifetime is equal to $0.1 \%$. The final values of $\tau^{\mathrm{CI}+\text { all }}$ for the 23 lowest-lying levels are listed in the last column of Table VI. The term designation for those levels are in the first column of Table VI. In Table VII we present results for other $E 1$ transitions for low-lying levels with smaller transition rates.

Unfortunately, we did not find any theoretical or experimental results to compare with our $A_{r}$ and $\tau$ values for the lowlying states listed in Table VI. Experimental measurements of lifetimes for six higher levels of $\mathrm{Th}^{2+}$ were performed by Biémont et al. [22] and are summarized in Table VIII. In order to calculate these lifetimes with our CI plus all-order method, we carried out extensive additional calculations to obtain the wave functions and corresponding electric dipole matrix elements for 112 levels with $0 \leqslant J \leqslant 6$. Using the calculated CI plus all-order electric dipole matrix elements and experimental energies [23], we obtain values for 1152

TABLE VI. Lifetimes $\tau^{\mathrm{Cl}+\text { all }}$ (in s), transition rates $A_{r}\left(\right.$ in $\left.^{-1}\right)$, and reduced matrix elements $Z^{\mathrm{Cl}+\text { all }}$ (in a.u.) for electric multipole ( $E 1$ and $E 2$ ) and magnetic multipole ( $M 1$ and $M 2$ ) transitions in the Ra-like $\mathrm{Th}^{2+}$ ion evaluated in the CI plus all approximation. Energies (in $\mathrm{cm}^{-1}$ ) are from Ref. [23]. The numbers in square brackets represent powers of 10.

| Transition |  |  | Energies ( $\mathrm{cm}^{-1}$ ) |  | (Å) | $Z^{\text {no RPA }}$ <br> (a.u.) | $\begin{gathered} Z^{\text {Cl+all }} \\ \text { (a.u.) } \end{gathered}$ | $\begin{gathered} A_{r}^{\mathrm{Cl}+\mathrm{all}} \\ \left(\mathrm{~s}^{-1}\right) \end{gathered}$ | $\tau^{\mathrm{Cl}+\mathrm{all}}$ <br> (s) |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Upper | Lower |  | Lower | Upper |  |  |  |  |  |
| $6 d^{23} F_{2}$ | $5 f 6 d^{3} H_{4}$ | M2 | 0.00 | 63.27 | 1580528 | 2.1134 | 1.4359 | 6.224[-19] | 1.607[18] |
| $5 f 6 d^{3} F_{2}$ | $5 f 6 d^{3} H_{4}$ | E2 | 0.00 | 510.76 | 195787 | 3.6338 | 3.1838 | 7.901[-9] | 1.266[8] |
| $5 f 7 s^{3} F_{3}$ | $6 d^{23} F_{2}$ | E1 | 63.27 | 2527.09 | 40587 | 1.1061 | 0.7668 | 2.546[3] | 3.928[-4] |
| $5 f 7 s^{3} F_{2}$ | $5 f 6 d^{3} H_{4}$ | E2 | 0.00 | 3181.50 | 31432 | 11.1410 | 10.6310 | 8.251[-4] | 3.038[2] |
|  | $5 f 7 s^{3} F_{3}$ | M1 | 2527.09 | 3181.50 | 152809 | 1.4806 | 1.4766 | 3.292[-3] |  |
| $5 f 6 d^{1} G_{4}$ | $5 f 6 d^{3} H_{4}$ | E2 | 0.00 | 3188.30 | 31365 | 5.4527 | 5.0061 | 1.028[-4] | 1.740[3] |
|  | $5 f 6 d^{3} F_{2}$ | E2 | 510.87 | 3188.30 | 37349 | 3.5556 | 3.4795 | 2.073[-5] |  |
|  | $5 f 7 s^{3} F_{3}$ | M1 | 2527.09 | 3188.50 | 151192 | 0.7239 | 0.7220 | 4.513[-4] |  |
| $6 d^{23} F_{3}$ | $5 f 6 d^{3} H_{4}$ | E1 | 0.00 | 4056.02 | 24655 | 0.0662 | 0.0214 | 8.841[0] | 1.235[-3] |
|  | $5 f 6 d^{3} F_{2}$ | E1 | 510.76 | 4056.02 | 28207 | 0.3400 | 0.2483 | 7.953[2] |  |
| $5 f 6 d^{3} H_{5}$ | $5 f 6 d^{3} H_{4}$ | M1 | 0.00 | 4489.64 | 22274 | 2.4360 | 2.4438 | $1.325[0]$ | 7.410[-1] |
|  | $5 f 6 d^{1} G_{4}$ | M1 | 3188.30 | 4489.64 | 76844 | 2.1292 | 2.1303 | 2.453[-2] |  |
| $5 f 6 d^{3} F_{3}$ | $6 d^{2}{ }^{3} F_{2}$ | E1 | 63.27 | 4826.83 | 20993 | 1.1113 | 0.7438 | 1.731[4] | 5.777[-5] |
| $6 d^{23} F_{2}$ | $5 f 7 s^{3} F_{3}$ | E1 | 2327.09 | 4676.43 | 42565 | 0.9413 | 0.5569 | 1.629[3] | 6.139[-4] |
| $5 f 6 d^{3} G_{3}$ | $6 d^{2}{ }^{3} F_{2}$ | E1 | 63.27 | 5061.54 | 20007 | 2.1735 | 1.4623 | 7.766[4] | 1.288[-5] |
| $6 d^{23} P_{0}$ | $6 d^{2}{ }^{3} F_{2}$ | E2 | 63.27 | 5090.06 | 19893 | 5.8665 | 5.3985 | 1.047[-2] | 9.555[1] |
| $6 d 7 s^{3} D_{1}$ | $5 f 6 d^{3} F_{2}$ | E1 | 510.76 | 5523.88 | 19948 | 0.7040 | 0.4896 | 2.040[4] | 4.159[-5] |
|  | $5 f 7 s^{3} F_{2}$ | E1 | 3181.50 | 5523.88 | 42692 | 1.1268 | 0.6477 | 3.642[3] |  |
| $5 f 6 d^{1} D_{2}$ | $6 d^{2}{ }^{3} F_{3}$ | E1 | 4056.02 | 6288.42 | 44795 | 0.1510 | 0.0727 | 2.379 [1] | 4.187[-2] |
| $5 f 7 s^{3} F_{4}$ | $6 d^{2}{ }^{3} F_{3}$ | E1 | 4056.02 | 6310.81 | 44350 | 0.7621 | 0.5514 | 7.849[2] | 1.273[-3] |
| $6 d^{23} F_{4}$ | $5 f 7 s^{3} F_{3}$ | E1 | 2527.09 | 6537.78 | 24933 | 0.2457 | 0.2126 | 6.568[2] | 1.131[-3] |
|  | $5 f 6 d^{3} F_{3}$ | E1 | 4826.83 | 6537.81 | 58446 | 0.3792 | 0.3273 | 1.208[2] |  |
|  | $5 f 6 d^{3} G_{3}$ | E1 | 5060.94 | 6537.81 | 67711 | 0.5535 | 0.3609 | 9.279[1] |  |
| $6 d 7 s^{3} D_{2}$ | $5 f 7 s^{3} F_{3}$ | E1 | 2527.09 | 7176.11 | 21510 | 0.9399 | 0.5919 | 1.426[4] | 6.720[-5] |
|  | $5 f 6 d^{3} F_{3}$ | $E 1$ | 4826.83 | 7176.11 | 42566 | 0.2150 | 0.3040 | 6.203[2] |  |
| $5 f 7 s^{3} F_{3}$ | $6 d^{2}{ }^{3} F_{2}$ | E1 | 63.27 | 7500.61 | 13446 | 0.1114 | 0.0762 | 6.907[2] | 3.182[-4] |
|  | $6 d^{2}{ }^{3} F_{2}$ | E1 | 4676.43 | 7500.61 | 35409 | 0.8405 | 0.6070 | $2.403[3]$ |  |
|  | $6 d^{2}{ }^{3} F_{4}$ | E1 | 6537.81 | 7500.61 | 103864 | 0.4824 | 0.4054 | 4.249[1] |  |
| $6 d^{2}{ }^{3} P_{1}$ | $5 f 6 d^{3} F_{2}$ | E1 | 510.76 | 7875.83 | 13578 | 0.1028 | 0.0840 | 1.902[3] | 4.898[-4] |
|  | $5 f 7 s^{3} F_{2}$ | E1 | 3181.50 | 7875.83 | 21302 | 0.0712 | 0.0446 | 1.392[2] |  |
| $5 f 6 d^{3} G_{4}$ | $6 d^{2}{ }^{3} F_{3}$ | $E 1$ | 4056.02 | 8141.75 | 24475 | 2.7360 | 1.8865 | 5.465[4] | 1.830[-5] |
| $6 d 7 s^{3} D_{3}$ | $5 f 6 d^{3} H_{4}$ | E1 | 0.00 | 9953.58 | 10047 | 0.2032 | 0.1630 | 7.579[3] | 2.666[-5] |
|  | $5 f 6 d^{1} G_{4}$ | E1 | 3188.50 | 9953.58 | 14782 | 0.4667 | 0.3538 | 1.122[4] |  |
|  | $5 f 7 s^{3} F_{4}$ | $E 1$ | 6310.81 | 9953.58 | 27452 | 1.6388 | 1.0983 | 1.687[4] |  |
| $6 d^{2}{ }^{3} P_{2}$ | $5 f 6 d^{3} F_{3}$ | E1 | 4826.83 | 10440.24 | 17814 | 0.1880 | 0.1106 | 8.772[2] | 1.982[-4] |
|  | $5 f 6 d^{3} G_{3}$ | $E 1$ | 5060.54 | 10440.24 | 18588 | 0.2160 | 0.1468 | $1.353[3]$ |  |
|  | $5 f 7 s^{3} F_{3}$ | $E 1$ | 7500.61 | 10440.24 | 34018 | 0.7666 | 0.5082 | 2.658[3] |  |
| $6 d^{2}{ }^{1} G_{4}$ | $5 f 6 d^{3} F_{3}$ | E1 | 4826.83 | 10542.90 | 17495 | 1.0369 | 0.7953 | $2.949[4]$ | $2.366[-5]$ |
|  | $5 f 6 d^{3} G_{3}$ | E1 | 5060.54 | 10542.90 | 18240 | 0.7825 | 0.5208 | 1.001[4] |  |
| $7 s^{2}{ }^{1} S_{0}$ | $6 d^{2}{ }^{3} F_{2}$ | E2 | 4676.43 | 11961.13 | 13727 | 7.0555 | 6.7483 | 1.046[-1] | 1.418[0] |
|  | $6 d^{2}{ }^{3} P_{1}$ | M1 | 7875.83 | 11961.13 | 24478 | 0.5484 | 0.5504 | 5.572[-1] |  |

transition rates $A_{r}$. These are summed over all transitions to determine the lifetimes for about 100 levels.

In Table VIII we present lifetimes $\tau^{\mathrm{CI}+\text { all }}$, transition rates $A_{r}$, and branching ratios of electric dipole transitions relevant to the comparison with the lifetime measurements of [22]. Energies in both tables are from the compilation of Ref. [23]. Only the dominant transitions are listed, but all transitions are included in the lifetime calculation.

In the two last columns of Table VIII we compare our CI plus all-order lifetimes with measurements [22].

We find excellent agreement between the CI plus allorder lifetimes and experimental values for the $5 f 7 p^{3} G_{4}$, $7 s 7 p^{3} P_{0}, 7 s 7 p^{3} P_{1}$, and $6 d 7 p^{3} F_{4}$ levels. There is a $15 \%$ difference for the lifetime of the $5 f^{23} P_{2}$ level. Several channels contribute significantly to this lifetime. The largest branching ratio for this level is 0.31 for the $5 f 6 d^{3} P_{2}-5 f^{2}{ }^{3} P_{2}$ transition. The branching ratios for the other five transitions shown in Table VIII add $56 \%$. An additional $13 \%$ comes from eight transitions that are not shown in Table VIII.

TABLE VII. Oscillator strengths $f$ and transition rates $A_{r}\left(\mathrm{in} \mathrm{s}^{-1}\right)$ for electric dipole transitions in the Ra-like $\mathrm{Th}^{2+}$ ion. Wavelengths (in $\AA$ ) from the compilation of Ref. [23] are listed for reference. The numbers in square brackets represent powers of 10 .

| Transition |  | Wavelength | $f^{\text {CI }+ \text { all }}$ | $A_{r}^{\text {CI+all }}$ |
| :--- | :---: | :---: | :---: | :---: | :---: |
| Lower | Upper | $(\AA)$ |  | $\left(\mathrm{s}^{-1}\right)$ |
| $5 f 6 d^{3} H_{4}$ | $6 d^{2}{ }^{3} G_{4}$ | 9485.6 | $3.733[-2]$ | $3.417[4]$ |
| $6 d^{2}{ }^{3} F_{2}$ | $5 f 6 d^{3} D_{1}$ | 12726.7 | $1.651[-2]$ | $4.532[4]$ |
| $5 f 7 s^{3} F_{3}$ | $6 d 7 s^{3} D_{3}$ | 13465.8 | $1.664[-2]$ | $1.250[4]$ |
| $5 f 6 d^{3} F_{2}$ | $6 d 7 s^{3} D_{2}$ | 15003.5 | $9.591[-3]$ | $1.137[4]$ |
| $5 f 6 d^{3} H_{4}$ | $6 d^{2}{ }^{3} F_{4}$ | 15296.1 | $6.768[-3]$ | $2.383[3]$ |
| $6 d^{2}{ }^{1} D_{2}$ | $5 f 6 d^{3} P_{1}$ | 15512.2 | $6.619[-3]$ | $1.223[4]$ |
| $6 d^{2}{ }^{3} F_{2}$ | $5 f 6 d^{1} D_{2}$ | 16064.4 | $1.080[-3]$ | $1.117[3]$ |
| $6 d^{2} F_{3}$ | $5 f 6 d^{3} D_{2}$ | 16327.7 | $3.118[-2]$ | $2.229[4]$ |
| $6 d^{2}{ }^{1} D_{2}$ | $5 f 6 d^{3} D_{3}$ | 16489.3 | $6.448[-2]$ | $4.520[4]$ |
| $6 d 7 s^{3} D_{1}$ | $5 f 6 d^{3} P_{0}$ | 17515.8 | $4.814[-3]$ | $3.488[4]$ |
| $6 d 7 s^{3} D_{1}$ | $5 f 6 d^{3} P_{1}$ | 17859.9 | $4.407[-3]$ | $1.024[4]$ |
| $6 d^{2} F_{2}$ | $5 f 6 d^{3} G_{3}$ | 20011.4 | $1.625[-1]$ | $7.737[4]$ |
| $6 d^{2}{ }^{3} F_{3}$ | $5 f 6 d^{3} F_{4}$ | 20307.0 | $9.844[-2]$ | $2.528[4]$ |
| $6 d^{2} F_{4} F_{4}$ | $5 f 6 d^{3} G_{5}$ | 21102.0 | $5.705[-1]$ | $8.633[4]$ |
| $6 d^{2}{ }^{3} F_{4}$ | $5 f 6 d^{3} D_{3}$ | 23791.1 | $1.139[-1]$ | $2.131[4]$ |
| $5 f 6 d^{3} F_{2}$ | $6 d^{21} D_{2}$ | 24006.2 | $8.657[-3]$ | $4.007[3]$ |
| $5 f 6 d^{3} D_{1}$ | $7 s^{2}{ }^{1} S_{0}$ | 24752.7 | $7.869[-4]$ | $2.857[3]$ |
| $5 f 6 d^{3} F_{2}$ | $6 d^{2}{ }^{3} F_{3}$ | 28207.2 | $3.317[-3]$ | $7.947[2]$ |
| $5 f 6 d^{1} G_{4}$ | $6 d^{2}{ }^{3} F_{4}$ | 29855.6 | $2.675[-2]$ | $2.473[3]$ |
| $6 d^{2}{ }^{3} F_{2}$ | $5 f 6 d^{3} F_{2}$ | 32070.0 | $1.603[-3]$ | $4.159[2]$ |
| $6 d^{2}{ }^{1} D_{2}$ | $5 f 7 s^{1} F_{3}$ | 35396.5 | $1.569[-2]$ | $2.386[3]$ |
| $5 f 7 s^{3} F_{3}$ | $6 d^{21} D_{2}$ | 46526.4 | $1.439[-2]$ | $1.266[3]$ |
| $6 d^{2}{ }^{1} D_{2}$ | $5 f 6 d^{1} D_{2}$ | 62035.6 | $1.691[-3]$ | $1.172[2]$ |
| $5 f 7 s^{3} F_{3}$ | $6 d^{23} F_{3}$ | 65405.7 | $1.286[-2]$ | $4.094[2]$ |
| $6 d^{2}{ }^{3} F_{2}$ | $5 f 6 d^{3} F_{2}$ | 223469.2 | $3.053[-3]$ | $1.637[1]$ |

## VI. CONCLUSION

In summary, a systematic study of the $\mathrm{Th}, \mathrm{Th}^{+}$, and $\mathrm{Th}^{2+}$ energies was carried out using the CI plus all-order approach. Excitation energies are compared with experimental [23] and available theoretical results [33]. Good agreement with experiment was found even for neutral Th owing to the all-order treatment of the dominant correlation corrections and sufficient saturation of the configuration space. The differences between the theoretical and experimental energies of neutral thorium did not exceed $2 \%$ for 35 out of 78 levels listed in Table I and only 9 energies differed with experiment for more than $5 \%$. These result show the success of the algorithm that we have developed for efficient selection of important configurations for tetravalent systems.

We explored the issue of accidentally vanishing $g$ factors, where the Landè formula gives $g=0$, such as for ${ }^{5} F_{1}$ terms. We identified a number of cases in which hyperfine $g$ factors may also vanish.

The recommended values are provided for multipole transition rates and lifetimes of low-lying levels in Ra-like $\mathrm{Th}^{2+}$. We expect these values to be accurate to a few percent for strong transitions based on our calculations in divalent alkaline-earth atoms [54]. To further verify the accuracy of our calculations, we compared our results with the only available experimental lifetimes [22] for higher $5 f 7 p^{3} G_{4}, 7 s 7 p^{3} P_{0}, 7 s 7 p^{3} P_{1}$, and $6 d 7 p^{3} F_{4}$ levels of $\mathrm{Th}^{2+}$ ion. This works demonstrates the ability to perform accurate calculations for Th and its ions for Th nuclear clock development and other applications.

TABLE VIII. Lifetimes (in ns), transition rates $A_{r}\left(\mathrm{in} \mathrm{s}^{-1}\right)$, and branching ratios of electric dipole transitions in Ra-like $\mathrm{Th}^{2+}$. Experimental lifetimes are taken from Ref. [22]. Levels (in $\mathrm{cm}^{-1}$ ) are from the experimental compilation of Ref. [23]. The numbers in square brackets represent powers of 10 .

| Transition |  | Levels ( $\mathrm{cm}^{-1}$ ) |  | $\lambda$ <br> (Å) | $Z^{\mathrm{CI}+\mathrm{all}}$ <br> (a.u.) | $\begin{gathered} A_{r}^{\mathrm{CI}+\mathrm{all}} \\ \left(\mathrm{~s}^{-1}\right) \end{gathered}$ | Branching ratio | $\begin{gathered} \tau^{\mathrm{CI}+\mathrm{all}} \\ (\mathrm{~ns}) \end{gathered}$ | $\begin{aligned} & \tau^{\text {Expt }} \\ & \text { (ns) } \end{aligned}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Upper | Lower | Upper | Lower |  |  |  |  |  |  |
| $5 f^{2}{ }^{3} P_{2}$ | $5 f 6 d^{3} F_{2}$ | 32867.27 | 510.76 | 3090.6 | 0.8137 | 9.09[6] | 0.19 | 21.2 | $25.8 \pm 1.5$ |
| $5 f^{2}{ }^{3} P_{2}$ | $5 f 6 d^{1} D_{2}$ | 32867.27 | 6288.42 | 3762.4 | 0.7188 | 3.93[6] | 0.08 |  |  |
| $5 f^{2}{ }^{3} P_{2}$ | $5 f 6 d^{3} D_{3}$ | 32867.27 | 10741.15 | 4519.5 | 1.0101 | 4.48[6] | 0.10 |  |  |
| $5 f^{2}{ }^{3} P_{2}$ | $5 f 6 d^{3} P_{1}$ | 32867.27 | 11123.18 | 4599.0 | 0.8850 | 3.26[6] | 0.07 |  |  |
| $5 f^{2}{ }^{3} P_{2}$ | $5 f 6 d^{3} P_{2}$ | 32867.27 | 13208.21 | 5086.7 | 2.1928 | 1.48[7] | 0.31 |  |  |
| $5 f^{2}{ }^{3} P_{2}$ | $5 f 6 d^{3} F_{3}$ | 32867.27 | 15453.41 | 5742.6 | 1.6403 | 5.76[6] | 0.12 |  |  |
| $5 f 7 p^{3} G_{4}$ | $5 f 6 d^{1} G_{4}$ | 38580.60 | 3188.30 | 2825.5 | 2.0580 | 4.23[7] | 0.10 | 2.41 | $2.7 \pm 0.2$ |
| $5 f 7 p^{3} G_{4}$ | $5 f 6 d^{3} H_{5}$ | 38580.60 | 4489.64 | 2933.3 | 3.1200 | 8.68[7] | 0.21 |  |  |
| $5 f 7 p^{3} G_{4}$ | $5 f 7 s^{3} F_{4}$ | 38580.60 | 6310.81 | 3098.9 | 3.8614 | 1.13[8] | 0.27 |  |  |
| $5 f 7 p^{3} G_{4}$ | $5 f 7 s^{1} F_{3}$ | 38580.60 | 7500.61 | 3217.5 | 3.2880 | 7.31[7] | 0.18 |  |  |
| $5 f 7 p^{3} G_{4}$ | $5 f 6 d^{3} F_{4}$ | 38580.60 | 8980.56 | 3378.4 | 2.5555 | 3.81[7] | 0.09 |  |  |
| $7 s 7 p^{3} P_{0}$ | $6 d 7 s^{3} D_{1}$ | 42259.71 | 5523.88 | 2722.1 | 1.2104 | 1.47[8] | 0.91 | 6.19 | $6.6 \pm 0.4$ |
| $7 s 7 p^{3} P_{0}$ | $6 d^{2}{ }^{3} P_{1}$ | 42259.71 | 7875.83 | 2908.3 | 0.4150 | 1.42[7] | 0.09 |  |  |
| $7 s 7 p^{3} P_{1}$ | $6 d^{2}{ }^{3} F_{2}$ | 45063.97 | 63.27 | 2222.2 | 1.0592 | 6.91[7] | 0.15 | 2.22 | $2.4 \pm 0.2$ |
| $7 s 7 p^{3} P_{1}$ | $6 d^{2}{ }^{1} D_{2}$ | 45063.97 | 4676.43 | 2476.0 | 2.2857 | 2.33[8] | 0.52 |  |  |
| $7 s 7 p^{3} P_{1}$ | $6 d 7 s^{3} D_{1}$ | 45063.97 | 5523.88 | 2529.1 | 0.7118 | 2.12[7] | 0.05 |  |  |
| $7 s 7 p^{3} P_{1}$ | $6 d 7 s^{3} D_{2}$ | 45063.97 | 7176.11 | 2639.4 | 1.4716 | 7.96[7] | 0.18 |  |  |
| $7 s 7 p^{3} P_{1}$ | $7 s^{2}{ }^{1} S_{0}$ | 45063.97 | 11961.13 | 3020.9 | 0.8471 | 1.76[7] | 0.04 |  |  |
| $6 d 7 p^{3} F_{4}$ | $6 d^{2}{ }^{3} F_{3}$ | 53052.47 | 4056.02 | 2041.0 | 0.7944 | 1.67[7] | 0.02 | 1.41 | $1.3 \pm 0.2$ |
| $6 d 7 p^{3} F_{4}$ | $6 d^{2}{ }^{3} F_{4}$ | 53052.47 | 6537.81 | 2149.9 | 2.3177 | 1.22[8] | 0.17 |  |  |
| $6 d 7 p^{3} F_{4}$ | $6 d 7 s^{3} D_{3}$ | 53052.47 | 9953.58 | 2320.2 | 5.3100 | 5.08[8] | 0.72 |  |  |
| $6 d 7 p^{3} F_{4}$ | $6 d^{2}{ }^{3} G_{4}$ | 53052.47 | 10542.90 | 2352.4 | 1.7656 | 5.39[7] | 0.08 |  |  |

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